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(71) Applicant (for all designated States except US): THE REGENTS OF THE UNIVERSITY OF CALIFORNIA [US/US]; 22nd floor, 300 Lakeside Drive, Oakland, CA 94612 (US).

(75) Inventors/Applicants (for US only): TSIEN, Roger, Y. [US/US]; 8535 Nottingham Place, La Jolia, CA 92037 (US/US); ZLOKARNIK, Gregor [DE/US]; 3329 Jemez Drive, San Diego, CA 92117 (US).

(74) Agent: STORELLA, John, R.; Townsend and Townsend and Crew, Steuart Street Tower, 20th floor, One Market Plaza, San Francisco, CA 94105 (US).

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Substrates for  $\beta$ -lactamase of general formula (I) in which one of X and Y is a fluorescent donor molety and the other is a quencher (which may or may not re-emit); R' is selected from the group consisting of H, lower (i.e., alkyl of 1 to about 5 carbon atoms) and (CH<sub>2</sub>)<sub>n</sub>OH, in which n is 0 or an integer from 1 to 5; R" is selected from the group consisting of H, physiologically acceptable metal and ammonium cations, -CHR<sup>2</sup>OCO(CH<sub>2</sub>)<sub>n</sub>CH<sub>3</sub>, -CHR<sup>2</sup>OCO(CH<sub>3</sub>)<sub>3</sub>, acylthiomethyl, alpha-benzyl, delta-butyrolactonyl, methoxycarbonyloxymethyl, phenyl, methylsulphinylmethyl, beta-morpholinoethyl, dialkylaminoethyl, dialkylaminocarbonyloxymethyl and aliphatic, in which R<sup>2</sup> is selected from the group consisting of H and lower alkyl; A is selected from the group consisting of S, O, SO, SO<sub>2</sub> and CH<sub>2</sub>; and Z' are linkers for the fluorescent donor and quencher moleties. Selected from the group consisting of S, O, SO, SO<sub>2</sub> and CH<sub>2</sub>; and Z' are linkers for the fluorescent donor and quencher moleties. Methods of assaying  $\beta$ -lactamase activity and monitoring expression in systems using  $\beta$ -lactamase as a reporter gene also are disclosed. Methods of assaying  $\beta$ -lactamase activity and monitoring expression in systems using  $\beta$ -lactamase as a reporter gene also are disclosed.

Applicants: Eric A. Althoff and Virginia W.

Cornish

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Exhibit 11

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# INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

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$$X$$
  $-z$   $R'$   $A$   $Z'$   $Y$   $R$ 

Substrates for β-lactamase of general formula (I) in which one of X and Y is a fluorescent donor moiety and the other is a quencher (which may or may not re-emit). R' is selected from the group consisting of H, lower (i.e., alkyl of 1 to about 5 carbon atoms) and (CH2)<sub>8</sub>OH, in which n is 0 or an integer from 1 to 5; R" is selected from the group consisting of 5 carbon atoms) and (CH2)<sub>8</sub>OH, in which n is 0 or an integer from 1 to 5; R" is selected from the group consisting of H, physiologically acceptable metal and ammonium cations, -CHR<sup>2</sup>OCOC(CH<sub>2</sub>)<sub>8</sub>CH<sub>3</sub>, -CHR<sup>2</sup>OCOC(CH<sub>3</sub>)<sub>3</sub>, acylinhomethyl, acylony-the bound do by hyperbolic metal and ammonium cations, about metallic m n, physicologically acceptance makes and antiferrometric canonic control of the property of the physicological selected from the group consisting of S, O, SO, SO<sub>2</sub> and CH<sub>2</sub>; and Z' are linkers for the fluorescent donor and quencher molectes. Methods of assaying  $\beta$ -lactamase activity and monitoring expression in systems using  $\beta$ -lactamase as a reporter gene also are disclosed.

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# SUBSTRATES FOR BETA-LACTAMASE AND USES THEREOF

### BACKGROUND OF THE INVENTION

The present invention relates generally to the fields of chemistry and biology. More particularly, the present invention relates to compositions and methods for use in measuring gene expression.

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gene's promoter. It takes advantage of molecular biology techniques, which allow one to put heterologous genes under the control of any promoter and introduce the construct into the genome of a mammalian cell [Gorman, C.M. et al., Mol. Cell Biol. 2: 1044-1051 (1982); Alam, J. and Cook, J.L., Anal.Biochem. 188: 245-254, (1990)]. Activation of the promoter induces the reporter gene as well as or instead of the endogenous gene. By design the reporter gene codes for a protein that can easily be detected and measured. Commonly it is an enzyme that converts a commercially available substrate into a product. This conversion is conveniently followed by either chromatography or direct optical measurement and allows for the quantification of the amount of enzyme produced.

Reporter genes are commercially available on a variety of plasmids for the study of gene regulation in a large variety of organisms [Alam and Cook, supra]. Promoters of interest can be inserted into multiple cloning sites provided for this purpose in front of the reporter gene on the plasmid [Rosenthal, N., Methods Enzymol. 152: 704-720 (1987); Shiau, A. and Smith, J.M., Gene 67: 295-299 (1988)]. Standard techniques are used to introduce these genes into a cell type or whole organism [e.g., as described in Sambrook, J., Fritsch, E.F. and Maniatis, T. Expression of cloned genes in cultured mammalian cells. In: Molecular Cloning, edited by Nolan, C. New York: Cold Spring Harbor Laboratory Press,

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1989]. Resistance markers provid d on the plasmid can then be used to select for successfully transf cted c lls.

Ease of use and the large signal amplification make this technique increasingly popular in the study of gene regulation. Every step in the cascade DNA --> RNA --> Enzyme --> Product --> Signal amplifies the next one in the sequence. The further down in the cascade one measures, the more signal one obtains.

In an ideal reporter gene assay, the reporter gene under the control of the promoter of interest is transfected into cells, either transiently or stably. Receptor activation leads to a change in enzyme levels via transcriptional and translational events. The amount of enzyme present can be measured via its enzymatic action on a substrate. The substrate is a small uncharged molecule that, when added to the extracellular solution, can penetrate the plasma membrane to encounter the enzyme. A charged molecule can also be employed, but the charges need to be masked by groups that will be cleaved by endogenous cellular enzymes (e.g., esters cleaved by cytoplasmic esterases).

For a variety of reasons, the use of substrates which exhibit changes in their fluorescence spectra upon interaction with an enzyme are particularly desirable. In some assays, the fluorogenic substrate is converted to a fluorescent product. Alternatively, the fluorescent substrate changes fluorescence properties upon conversion at the reporter enzyme. The product should be very fluorescent to obtain maximal signal, and very polar, to stay trapped inside the cell.

To achieve the highest possible sensitivity in a reporter assay one has to maximize the amount of signal generated by a single reporter enzyme. An optimal enzyme will convert 10<sup>5</sup> substrate molecules per second under saturating conditions [Stryer, L. Introduction to enzymes. In: Biochemistry, New York: W. H. Freeman and company, 1981, pp. 103-134]. β-Lactamases will cleave about 10<sup>3</sup> molecules of their favorite substrates per second [Chang, Y.H. et al., Proc.Natl.Acad.Sci.USA 87: 2823-2827 (1990)]. Using a

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fluorogenic substrate ne can obtain up to 10' photons per fluorescent product produc d, depending on the type of dye used, when exciting with light f th appropriate wavelength. The signal terminates with the bl aching of the fluorophore [Tsien, R.Y. and Waggoner, A.S. Fluorophores for confocal microscopy: Photophysics and photochemistry. In: Handbook of Biological Confocal Microscopy, edited by Pawley, J.B. Plenum Publishing Corporation, 1990, pp. 169-178]. These numbers illustrate the theoretical magnitude of signal obtainable in this type of measurement. In practice a minute fraction of the photons generated will be detected, but this holds true for fluorescence, bioluminescence or chemiluminescence. good fluorogenic substrate for a reporter enzyme has to have a high turnover at the enzyme in addition to good optical properties such as high extinction and high fluorescence quantum yield.

### SUMMARY OF THE INVENTION

It is an object of the present invention to provide β-lactamase substrate compounds. It is a further object of the invention to provide membrane-permeant compounds. The membrane-permeant compounds may be transformed into substantially membrane-impermeant compounds.

Another object of the invention is to provide  $\beta$ lactamase reporter genes. A further object of the present
invention is to create cells containing the  $\beta$ -lactamase
reporter genes functionally linked to a promotor such that
when the promotor is turned on, the reporter gene will be
expressed. Expression of the  $\beta$ -lactamase is measured with the  $\beta$ -lactamase substrates which emit light after hydrolysis by
the  $\beta$ -lactamase.

A further object of the invention is to use the  $\beta$ lactamase reporter genes in cells and the  $\beta$ -lactamase
substrate compounds of the present invention to screen for
biochemical activity.

In accordance with the present invention, fluorogenic substrates are provided of the general f rmula I

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wherein:

one of X and Y is a fluorescent donor moiety or a membrane-permeant derivative thereof, and the other is a quencher moiety, an acceptor fluorophore moiety or a membranepermeant derivative thereof;

R' is selected from the group consisting of H, lower alkyl,  $(CH_2)_nOH$ ,  $(CH_2)_nCOOR^*$ , and =NOJ, in which n is 0 or an integer from 1 to 5 and J is H, Me, CH2COOH, CHMeCOOH, and CMe2COOH;

R" is selected from the group consisting of H, physiologically acceptable metal and ammonium cations, - $CHR^{2}OCO(CH_{2})_{n}CH_{1}$ ,  $-CHR^{2}OCOC(CH_{3})_{3}$ , acylthiomethyl, acyloxy-alphabenzyl, delta-butyrolactonyl, methoxycarbonyloxymethyl, phenyl, methylsulphinylmethyl, betamorpholinoethyl, dialkylaminoethyl, dialkylaminocarbonyloxymethyl, in which R2 is selected from the group consisting of H and lower alkyl; A is selected from the group consisting of S, O, SO,

SO, and CH2; Z' is a linker for X; and

Z" is a linker for Y.

In another aspect, this invention provides methods for determining whether a sample contains  $\beta$ -lactamase activity. The methods involve contacting the sample with a compound substrate of the invention, which exhibits fluorescence resonance energy transfer when the compound is excited; exciting the compound; and determining the degree of fluorescence resonance energy transfer in the sample. A

degree of fluorescence resonance en rgy transfer that is lower than an expected amount indicates the presence of  $\beta$ -lactamase activity. One embodiment of this method is for determining the amount of an enzyme in a sample. According to this method, determining the degree f fluor scence resonance energy transfer in the sample comprises determining the degree at a first and second time after contacting the sample with the substrate, and determining the difference in the degree of fluorescence resonance energy transfer. The difference in the degree of fluorescence resonance energy transfer reflects the amount of enzyme in the sample.

In another aspect, this invention provides recombinant nucleic acid molecule comprising expression control sequences adapted for function in a vertebrate cell and operably linked to a nucleotide sequence coding for the expression of a  $\beta$ -lactamase. It also provides recombinant nucleic acid molecules comprising expression control sequences adapted for function in a eukaryotic cell and operably linked to a nucleotide sequence coding for the expression of a cytosolic  $\beta$ -lactamase. In certain embodiments, the invention is directed to mammalian host cells transfected with these recombinant nucleic acid molecules.

In another aspect, this invention provides methods for determining the amount of  $\beta$ -lactamase activity in a cell. The methods involve providing a host cell transfected with a recombinant nucleic acid molecule comprising expression control sequences operatively linked to nucleic acid sequences coding for the expression of a  $\beta$ -lactamase; contacting a sample comprising the cell or an extract of the cell with a substrate for  $\beta$ -lactamase; and determining the amount of substrate cleaved, whereby the amount of substrate cleaved is related to the amount of  $\beta$ -lactamase activity.

In another aspect, this invention provides methods for monitoring the expression of a gene operably linked to a set of expression control sequences. The methods involve providing a host cell transfected with a recombinant nucleic acid molecule comprising expression control sequences operatively linked to nucleic acid sequences coding for the

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expression of a  $\beta$ -lactamase, except if th eukaryot is a fungus, wher in the  $\beta$ -lactamase is a cytosolic  $\beta$ -lactamase; c ntacting a sample comprising the cell or an extract of the c ll or conditioned medium with a substrate for  $\beta$ -lactamase; and determining the amount of substrate cl aved. The amount of substrate cleaved is related to the amount of  $\beta$ -lactamase activity.

In another aspect, this invention provides methods for determining whether a test compound alters the expression of a gene operably linked to a set of expression control sequences. The methods involve providing a cell transfected 10 with a recombinant nucleic acid construct comprising the expression control sequences operably linked to nucleic acid sequences coding for the expression of a  $\beta$ -lactamase except if the eukaryote is a fungus, wherein the  $\beta$ -lactamase is a cytosolic  $\beta$ -lactamase; contacting the cell with the test 15 compound; contacting a sample comprising the cell or an extract of the cell with a  $\beta$ -lactamase substrate; and determining the amount of substrate cleaved, whereby the amount of substrate cleaved is related to the amount of  $\beta$ lactamase activity. In one embodiment of the methods, the 2,0 substrate is a compound of this invention. The step of determining the amount of substrate cleaved comprises exciting the compound; and determining the degree of fluorescence resonance energy transfer in the sample. A degree of fluorescence resonance energy transfer that is lower than an 25 expected amount indicates the presence of  $\beta$ -lactamase activity.

In another aspect, this invention provides methods of clonal selection comprising providing cells transfected with a recombinant nucleic acid molecule comprising the expression control sequences operably linked to nucleic acid sequences coding for the expression of a cytosolic  $\beta$ -lactamase; contacting the cells with a substance that activates or inhibits the activation of the expression control sequences; contacting the cells with a compound of claim 9 which is converted into a substrate; determining whether substrate is cleaved within each individual cell, whereby

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cleavage reflects  $\beta$ -lactamase activity; selecting and propagating those cells with a selected level of  $\beta$ -lactamas activity. In a further embodiment, the method further involves culturing select d cells in the absence of activat r for a time sufficient for cleaved substrate to be substantially lost from the cells and for  $\beta$ -lactamase l vels to return to unactivated levels; incubating the selected cells with a compound of claim 9 which is converted into a substrate; and selecting cells that have not substantially cleaved the substrate.

# BRIEF DESCRIPTION OF THE DRAWINGS

Figs. 1(a) and 1(b) illustrate the emission spectra for the fluorescein (a) and rhodamine (b) components of compound 11 (Example 1) before and after ß-lactamase cleavage of the ß-lactam ring;

Fig. 2 illustrates the emission spectrum of compound 17 before and after ß-lactamase cleavage of the ß-lactam ring;
Fig. 3 illustrates the emission spectrum of compound 22 before and after ß-lactamase cleavage of the ß-lactam ring;
Fig. 4 illustrates the emission spectrum of compound 25 before and after ß-lactamase cleavage of the ß-lactam ring;
Fig. 5 illustrates the emission spectrum of compound CCF2 before and after ß-lactamase cleavage of the ß-lactam

ring; and

Fig. 6 illustrates the emission spectrum of compound

CCF1 before and after ß-lactamase cleavage of the ß-lactam

ring

Fig. 7A presents a table describing various nucleotide and amino acid sequences useful in the invention.

Fig. 7B-C depicts Sequence 1, the nucleotide and deduced amino acid sequence of E. coli RTEM as modified by Kadonaga et al (1984).

Fig. 7D-E depicts Sequence 2, the nucleotide and deduced amino acid sequence of Wild-type secreted RTEM enzyme with Ser2-Arg, Ala23-Gly.

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Fig. 7F-G depicts Sequence 3, the nucleotide and deduced amino acid sequence of RTEM enzyme with  $\beta$ -globin upstream 1 ader, mammalian Kozak s quence, replacement of signal sequ no by Met Gly.

Fig. 7H-I depicts Sequence 4, the nucleotide and deduced amino acid sequence of RTEM  $\beta$ -lactamase with mammalian Kozak sequence and replacement of signal sequence by Met Asp.

Fig. 7J-K depicts Sequence 5, the nucleotide and deduced amino acid sequence of Bacillus licheniformis  $\beta$ lactamase with signal sequence replaced. 10

### DESCRIPTION OF THE INVENTION

### **DEFINITIONS**

In accordance with the present invention and as used herein, the following terms are defined with the following 15 meanings, unless stated otherwise.

The term "fluorescent donor moiety" refers the radical of a fluorogenic compound which can absorb energy and is capable of transferring the energy to another fluorogenic molecule or part of a compound. Suitable donor fluorogenic molecules include, but are not limited to, coumarins and related dyes xanthene dyes such as fluoresceins, rhodols, and rhodamines, rescrufins, cyanine dyes, bimanes, acridines, isoindoles, dansyl dyes, aminophthalic hydrazides such as luminol and isoluminol derivatives, aminophthalimides, aminonaphthalimides, aminobenzofurans, aminoquinolines, dicyanohydroquinones, and europium and terbium complexes and related compounds.

The term "quencher" refers to a chromophoric molecule or part of a compound which is capable of reducing the emission from a fluorescent donor when attached to the donor. Quenching may occur by any of several mechanisms including fluorescence resonance energy transfer, photoinduced electron transfer, paramagnetic enhancement of intersystem crossing, Dexter exchange coupling, and exciton coupling such as the formation of dark complexes. The term "acceptor" as used herein refers to a quencher which operates via fluorescence resonance energy transfer. Many acceptors can re-

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emit the transferr d en rgy as fluorescenc. Examples include coumarins and related fluorophores, xanthenes such as fluoresceins, rhodols, and rhodamines, resorufins, cyanines, difluoroboradiazaindacenes, and phthalocyanin s. Other chemical classes of acceptors generally do not r -emit the transferred energy. Examples include indigos, benzoquinones, anthraquinones, azo compounds, nitro compounds, indoanilines, di- and triphenylmethanes.

The term "dye" refers to a molecule or part of a compound which absorbs specific frequencies of light, including but not limited to ultraviolet light. The terms "dye" and "chromophore" are synonymous.

The term "fluorophore" refers to chromophore which fluoresces.

The term "membrane-permeant derivative" means a chemical derivative of a compound of general formula wherein at least one of X and Y contains at least one acylated aromatic hydroxyl, acylated amine, or alkylated aromatic hydroxyl wherein the acyl group contains 1 to 5 carbon atoms and wherein the alkyl group is selected from the group consisting of -CH2OC(0)alk, -CH2SC(0)alk, -CH2OC(0)Oalk, lower acyloxy-alpha-benzyl, and deltabutyrolactonyl; wherein alk is lower alkyl of 1 to 4 carbon atoms. These derivatives are made better able to cross cell membranes, i.e. membrane permeant, because hydrophilic groups are masked to provide more hydrophobic derivatives. Also, the masking groups are designed to be cleaved from the fluorogenic substrate within the cell to generate the derived substrate intracellularly. Because the substrate is more hydrophilic than the membrane permeant derivative it is now trapped within the cells.

The term "alkyl" refers to straight, branched, and cyclic aliphatic groups of 1 to 8 carbon atoms, preferably 1 to 6 carbon atoms, and most preferably 1 to 4 carbon atoms. The term "lower alkyl" refers to straight and branched chain alkyl groups of 1 to 4 carbon atoms.

The term "aliphatic" refers to saturated and unsaturated alkyl groups of 1 to 10 carbon atoms, preferably 1 to 6 carbon atoms, and most preferably 1 to 4 carbon atoms.

F-11-070,----

 $\beta$ -Lactamases are nearly optimal enzymes in respect SUBSTRATES to th ir almost diffusion-controlled catalysis of  $\beta$ -lactam hydrolysis [Christensen, H. et al., Biochem. J. 266: 853-861 (1990)]. Upon examination of the other properties of this class of enzymes, it was determined that they wer suited to the task of an intracellular reporter enzyme. They cleave the  $\beta$ -lactam ring of  $\beta$ -lactam antibiotics, such as penicillins and cephalosporins, generating new charged moieties in the process [O'Callaghan, C.H. et al., Antimicrob.Agents.Chemother. B: 57-63, (1968); Stratton, C.W., J. Antimicrob. Chemother. 22. Suppl. A: 23-35 (1988)]. A first generation cephalosporin is illustrated below, left, with the arrow pointing to the site of cleavage by  $\beta$ -lactamase. The free amino group thus generated (middle structure below) donates electron density through the vinyl group to promote irreversible cleavage of a nucleofugal group  $R_2$  from the 3'-position.  $R_2$  is thus free to diffuse away from the  $R_1$ -cephalosporin conjugate (right-hand structure below).

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 $\beta$ -Lactamases are a class of enzymes that have been very well characterized because of their clinical relevance in making bacteria resistant to  $\beta$ -lactam antibiotics [Waley, 30 S.G., Sci. Prog. 72: 579-597 (1988); Richmond, M.H. et al., Ann. N.Y. Acad. Sci. 182: 243-257 (1971)]. Most  $\beta$ -lactamases have been cloned and their amino acid sequence determined [see, e.g., Ambler, R.P., Phil. Trans. R. Soc. Lond. [Ser.B.] 35 289: 321-331 (1980)].

A gene encoding  $\beta$ -lactamase is known to molecular biologists as the ampicillin resistance gene (Ampr) and is

commonly us d to select for successfully transduced bacteria [Castagnoli, L. et al., Genet.Res. 40: 217-231 (1982)]; clones thereof are almost universally available. The enzyme catalyzes the hydrolysis f a β-lactam ring and will not accept peptides or protein substrates [Pratt, R.F. and Govardhan, C.P., Proc. Natl. Acad. Sci.USA 81: 1302-1306 (1984); Murphy, B.P. and Pratt, R.F., Biochemistry 30: 3640-3649 (1991)]. The kinetics of this reaction is well understood and there is no product inhibition [Bush, K. and Sykes, R.B., Antimicrob. Agents. Chemother. 30: 6-10 (1986); Christensen et al. (1990), supra). The enzyme substrates are less polar than the products.

The carboxyl group in the substrate can be easily masked by an acetoxymethyl ester [Jansen, A.B.A. and Russell, T.J., J.Chem.Soc. 2127-2132, (1965); Daehne, W. et al., J.Med.Chem. 11: 607-612 (1970)], which is readily cleaved by endogenous mammalian intracellular esterases. Conversion by these esterases followed by the  $\beta$ -lactam cleavage by  $\beta$ -lactamase generates two negative charges and a tertiary amine, which protonates. To date, there has been no report of a fluorogenic substrate with the appropriate properties, but multiple chromogenic substrates of different design have been reported and are commercially available [Jones, R.N. et al., J.Clin.Microbiol. 15: 677-683 (1982); Jones, R.N. et al., J.Clin.Microbiol. 15: 954-958 (1982); O'Callaghan, C.H. et al., Antimicrob.Agents.Chemother. 1: 283-288 (1972)].

and characterized, all of which would be suitable for use in accordance with the present invention. Initially, β-lactamases were divided into different classes (I through V) on the basis of their substrate and inhibitor profiles and their molecular weight [Richmond, M.H. and Sykes, R.B., Adv.Microb.Physiol. 2: 31-88 (1973)]. More recently, a classification system based on amino acid and nucleotide sequence has been introduced [Ambler, R.P., Phil, Trans. R. Soc. Lond. [Ser.B.] 289: 321-331 (1980)]. Class A β-lactamases possess a serine in the active site and have an approximate weight of 29kd. This class contains the plasmid-

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mediat d TEM  $\beta$ -lactamases such as the RTEM enzym of pBR322. Class B  $\beta$ -lactamases have an active-site zinc bound to a cysteine residu. Class C enzymes have an active site serine and a molecular weight of approximately 39kd, but have no amino acid homology to the class A enzymes.

The coding region of an exemplary ß-lactamase employed in the reporter gene assays described herein is indicated in SEQ ID NO:1 (nucleic acid sequence) and SEQ ID NO:2 (amino acid sequence). The pTG2del1 containing this sequence has been described [Kadonaga, J.T. et al., J.Biol.Chem. 259: 2149-2154 (1984)]. The entire coding sequence of wildtype pBR322 ß-lactamase has also been published [Sutcliffe, J.G., Proc.Natl.Acad.Sci.USA 75: 3737-3741 (1978)]. As would be readily apparent to those skilled in the field, this and other comparable sequences for peptides having ß-lactamase activity would be equally suitable for use in accordance with the present invention. The ß-lactamase reporter gene is employed in an assay system in a manner well known per se for the use of reporter genes (for example, in the form of a suitable plasmid vector).

In conjunction with a suitable ß-lactamase, there are employed in accordance with the present invention fluorogenic substrates of the general formula I

in which one of X and Y is a fluorescent donor moiety and the other is a quencher (which may or may not re-emit); R' is selected from the group consisting of H, lower alkyl, (CH<sub>2</sub>)<sub>n</sub>OH, (CH<sub>2</sub>)<sub>n</sub>COOR\*, and =NOJ, in which n is 0 or an integer from 1 to 5 and J is H, Me, CH<sub>2</sub>COOH, CHMeCOOH, and CMe<sub>2</sub>COOH; R\* is selected from the group consisting of H, physiologically acceptable metal and ammonium cations, -CHR<sup>2</sup>OCO(CH<sub>2</sub>)<sub>n</sub>CH<sub>3</sub>, -CHR<sup>2</sup>OCO(CH<sub>3</sub>), acylthiomethyl, acyloxy-alpha-benzyl, delta-

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butyrolactonyl, methoxycarbonyloxymethyl, phenyl, methylsulphinylmethyl, beta-morpholinoethyl, dialkylaminoethyl, and dialkylaminocarbonyloxymethyl, in which R<sup>2</sup> is selected from the group consisting of H and lower alkyl; A is select d from th group consisting f S, O, SO, SO<sub>2</sub> and CH<sub>2</sub>; and Z' and Z" are linkers for the fluorescent donor and quemcher moieties.

The linkers Z' and Z' serve the purpose of attaching the fluorescent donor and quencher moieties to the cephalosporin-derived backbone, and may facilitate the synthesis of the compounds of general formula I. In general formula I, Z' may represent a direct bond to the backbone; alternatively, suitable linkers for use as Z' include, but are not limited to, the following: -(CH<sub>2</sub>)<sub>a</sub>CONR<sup>2</sup>(CH<sub>2</sub>)<sub>a</sub>-, -

15 (CH<sub>2</sub>) ,NR<sup>2</sup>CO(CH<sub>2</sub>) ,-,

- $(CH_2)_nNR^3CONR^2(CH_2)_n$  ,  $(CH_2)_nNR^3CSNR^2(CH_2)_n$  ,
- $(CH_2)_n CONR^3 (CH_2)_p CONR^2 (CH_2)_n , (CH_2)_n ,$ 
  - $-(CH_{2})_{n}NR^{3}CO(CH_{2})_{p}S(CH_{2})_{n}-, -(CH_{2})_{n}S(CH_{2})_{n}-, -(CH_{2})_{n}O(CH_{3})_{n}-,$
  - $-(CH_2)_nNR^2(CH_2)_n-, -(CH_2)_nSO_2NR^2(CH_2)_n-, -(CH_2)_nCO_2(CH_2)_n-,$

wherein R<sup>2</sup> and n are as previously defined; R<sup>3</sup> is selected from the group consisting of hydrogen and lower alkyl; and each of m and p is independently selected from the group consisting of 0 and integers from 1 to 4. Especially preferred are Z' groups such where n and m are 0. Also particularly preferred are such Z' groups where R<sup>2</sup> is H.

Suitable linkers Z\* for the Y moiety include, but are not limited to, a direct bond to a heteroatom (e.g., O, N or S) in the dye's chromophore or the following:  $-O(CH_2)_n-, -S(CH_2)_n-, -NR^2(CH_2)_n-, -N^*R^2_2(CH_2)_n-, -OCONR^2(CH_2)_n-, -O_2C(CH_2)_n-, -SCSNR^2(CH_2)_n-, -SCSO(CH_2)_n-, -S(CH_2)_nCONR^2(CH_2)_n, -SCSO(CH_2)_n-, -S(CH_2)_nNR^2CO(CH_2)_n, and$ 

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in which  $R^2$ , n and m are as previously defined; and m is an integer from 0 to 4. Particularly preferred Z\* groups are -Especially pref rred is H.

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Preferred R' groups include H and methyl.

Particularly preferred is H. Preferred R" gr ups include H and acetoxymethyl. A preferred R2 group is H. A preferred A 5 group is -S-.

In a preferred aspect, the compounds of the present invention are membrane-permeant. Particularly preferred are such compounds wherein at least one of X and Y contains at least one acylated aromatic hydroxyl, acylated amine, or alkylated aromatic hydroxyl wherein the acyl group contains 1 to 5 carbon atoms and wherein the alkyl group is selected from the group consisting of -CH2OC(O)alk, -CH2SC(O)alk, -

CH<sub>2</sub>OC(O)Oalk, lower acyloxy-alpha-benzyl, and deltabutyrolactonyl, wherein alk is lower alkyl of 1 to 4 carbon 15 atoms. Particularly preferred are such compounds where at least one of X and Y contains at least one acylated aromatic hydroxy, wherein the acyl group is either acetyl, n-propionyl, -or-n-butyryl. Also particularly preferred are such compounds wherein at least one of X and Y contains an acetoxy methyl 20

group on an aromatic hydroxyl group.

In another preferred aspect, the quencher or acceptor is a fluorescein, rhodol, or rhodamin of formulae VIII-XII. Preferred are such compounds where the donor is a fluorescein of formula VIII and the quencher or acceptor is a rhodol or rhodamine of formulae VIII-XII. Also preferred are such compounds where the donor is a fluorescein of formula VIII and the quencher or acceptor is a tetrahalo fluorescein of formula VIII in which Ra, Rb, Rc, and Rd are independently Br or Cl. Also preferred are such compounds where the quencher or acceptor is a rhodol of formulae VIII, IX, and XI. Another preferred group of such compounds are those where the quencher or acceptor is a rhodamine of formulae VIII, X, and

In a another preferred aspect, the donor is a XII. coumarin of formulae II-VII and the quencher/acceptor is a fluorescein, rhodol, or rhodamine of formulae VIII-XII, XLVII,

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or XLVII, and membrane-permeant fluorogenic derivativ s thereof. Particularly preferred are such compounds with a fluor scein quencher/accept r of formula VIII. Especially preferred are such compounds where the coumarin is 7hydroxycoumarin or 7-hydroxy-6-chlorocoumarin and th fluorescein acceptor is fluorescein or dichlorofluorescein.

As would readily be appreciated by those skilled in the art, the efficiency of fluorescence resonance energy transfer depends on the fluorescence quantum yield of the donor fluorophore, the donor-acceptor distance and the overlap integral of donor fluorescence emission and acceptor absorption. The energy transfer is most efficient when a donor fluorophore with high fluorescence quantum yield (preferably, one approaching 100%) is paired with an acceptor with a large extinction coefficient at wavelengths coinciding with the emission of the donor. The dependence of fluorescence energy transfer on the above parameters has been reported [Forster, T. (1948) Ann. Physik 2: 55-75; Lakowicz, J.R., Principles of Fluorescence Spectroscopy, New York: Plenum Press (1983); Herman, B., Resonance energy transfer microscopy, in: Fluorescence Microscopy of Living Cells in Culture, Part B, Methods in Cell Biology, Vol 30, ed. Taylor, D.L. & Wang, Y.-L., San Diego: Academic Press (1989), pp. 219-243; Turro, N.J., Modern Molecular Photochemistry, Menlo Part: Benjamin/Cummings Publishing Co., Inc. (1978), pp. 296-361), and tables of spectral overlap integrals are readily available to those working in the field [for example, Berlman, I.B. Energy transfer parameters of aromatic compounds, Academic Press, New York and London (1973)]. The distance between donor fluorophore and acceptor dye at which fluorescence resonance energy transfer (FRET) occurs with 50% efficiency is termed R, and can be calculated from the spectral overlap integrals. For the donor-acceptor pair fluorescein tetramethyl rhodamine which is frequently used for distance measurement in proteins, this distance  $R_{\text{o}}$  is around 50-70 Å [dos Remedios, C.G. et al. (1987) J. Muscle Research and Cell 35 Motility 8:97-117]. The distance at which the energy transfer in this pair exceeds 90% is about 45 Å. When attached to the

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cephalosporin backbone the distances between donors and acceptors ar in the range of 10 Å to 20 Å, depending on the linkers used and the size of the chromophores. For a distance of 20 Å, a chromophore pair will have to have a calculated  $R_0$  of larger than 30 Å for 90% of the donors to transfer their energy to the acceptor, resulting in better than 90% quenching of the donor fluorescence. Cleavage of such a cephalosporin by  $\beta$ -lactamase relieves quenching and produces an increase in donor fluorescence efficiency in excess of tenfold.

Accordingly, it is apparent that identification of appropriate donor-acceptor pairs for use as taught herein in accordance with the present invention would be essentially routing to one skilled in the art.

To measure  $\beta$ -lactamase activity in the cytoplasm of living cells, smaller molecular weight chromophores as hereinafter described are in general preferred over larger ones as substrate delivery becomes a problem for larger compounds. Large molecules, especially those over about 1200 daltons, also tend to bind more avidly to cellular constituents than small ones, thereby removing at least some of them from access and cleavage by  $\beta$ -lactamase.

known to those skilled in the art. Generic structures of particular classes of chromophores suitable for use as X and Y are provided below. Compounds of general formulas II - XXXIV are exemplary of fluorophores which serve as the basis for particularly suitable donor moieties in the compounds of general formula I. Suitable chromophores for use as the basis of acceptor moieties in the compounds of general formula include, but are not limited to, compounds of general formulas II - LIV. Chromophores of general formulae XXXV-LIV usually do not re-emit efficiently.

# Coumarins and related dyes

# Xanthene dyes (including fluoresceins, rhodols and rhodamines)

### Cyanine dyes

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10 XVI ---15 XV

XVII

# Difluorchoradiazaindacene dyes

25 XVIII

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### **Bimanes**

### Acridines

### <u>Isoindoles</u>

SRh
NRk
XXIII
XXIV

### Dansyl dyes

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# Aminophthalic hydrazides (luminol and isoluminol derivatives)

RGRHN ON NH NH NH XXVIII

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**Aminophthalimides** 

10 Aminonaphthalimides

20 Aminoquinolines

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Aminobenzofurans

Dicyanohydroquinones

Indigo dyes

Anthraquinone dyes

Polymethine dyes

15 Nitro dyes and cyano derivatives

Ouinone dyes

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### Xanthene dyes

Dicyanovinyl and tricyanovinyl dyes

Indoaniline dyes (ninhydrin derivatives)

Di- and triphenylmethane dyes

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In pref rred emb dim nts f the compounds of general formulas II - LVI:

each of a and a' is independently H or an attachment point (i.e., a location at which the dye moiety is attached to the core structure of general formula I;

E is selected from the group consisting of H, OH,

ORk and NRSRh; G is selected from the group consisting of O and N'Rg'Rh';

each of L and L' is independently selected from the group consisting of CH and N;

M is selected from the group consisting of H, Mg, Al, Si, Zn, and Cu;

Q is selected from the group consisting of O, S, C(CH<sub>1</sub>)<sub>2</sub> and NR<sup>9</sup>;

Q' is selected from the group consisting of O, CH2, C(CH<sub>3</sub>)<sub>2</sub>, NR<sup>k</sup> and SO<sub>2</sub>;

T is selected from the group consisting of O and NRk;

each of W and W' is selected from the group consisting of O, S, Se and NH;

each of R<sup>a</sup>, R<sup>b</sup>, R<sup>c</sup> and R<sup>d</sup> is independently selected from the group consisting of an attachment point, H, halogen and lower alkyl;

R° is selected from the group consisting of an attachment point, H, lower alkyl, (CH2) CO2H,  $(CH_2)_nCHaCO_2H$ ,  $CHa(CH_2)_nCO_2H$ ,  $(CH_2)_nCOa$ , CH=CHCOa,

each of R', R', R', R', Rh, Rh and Rk is independently selected from the group consisting of an attachment point, H, lower alkyl and CH2(CH2) a;

R' is selected from the group consisting of an attachment point, H, halogen, lower alkyl, CN, CF,, phenyl, CO2H and CONR9'Rh';

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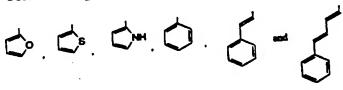
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R<sup>j</sup> is select d from the group consisting of an attachment point, H, halog n, lower alkyl, CN, CF<sub>3</sub>, phenyl, CH<sub>2</sub>CO<sub>2</sub>H, CH<sub>2</sub>CONR<sup>g</sup>'R<sup>h</sup>';

each of  $R^1$  and  $R^r$  is independently selected from the group consisting of an attachment point, H, lower alkyl,



each of  $R^n$ ,  $R^n$ ,  $R^p$  and  $R^q$  is independently selected from the group consisting of an attachment point, H, lower alkyl and phenyl;

R° is selected from the group consisting of an attachment point, H and lower alkyl;

each of  $R^s$  and  $R^t$  is independently selected from the group consisting of an attachment point, H, halogen, lower alkyl and  $OR^t$ ;

each of R" and R" is independently selected from the group consisting of an attachment point, H, halogen, CN and NO<sub>2</sub>;

each of R" is independently selected from the group—consisting of an attachment point, H, COO, SO, and PO, 2.

Ln is selected from the group consisting of  $Eu^{3+}$ ,  $Ln^{3+}$ , and  $Sm^{3+}$ ;

Chel is a polydentate chelator with at least six and preferably eight to ten donor atoms that can face into a cavity of diameter between 4 and 6 angstroms, which may or may not be macrocyclic, which includes a chromophore absorbing between 300 and 400 nm, and which includes an attachment point through which Chel can be conjugated to Z' or Z". A suitable Chel moiety is a europium tris-(bipyridine) cryptands. In the anthraquinone chromophores of general formula XXXIX, each of positions 1-8 may carry a substituent H or E, or serve as an attachment point.

Europium tris-(bipyridine)cryptand donors may be suitably paired with acceptors of the formulae XV-XVII, XXXVI, XLVI-XLVII, LIV, and LVI. Terbium tris-(bipyridine) cryptand

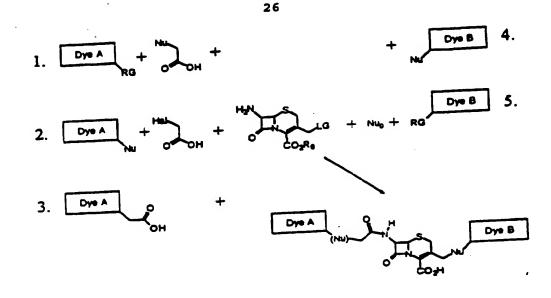
donors may be suitably paired with acceptors of the formula VIII-XVIII, XXXVI-XLI, and XLV-LIV, and LVI.

The Europium tris-(bipyridine) cryptand/ phtalocyanines donor/acceptor pair may be of particular interest when it is desirable to measure  $\beta$ -lactamase activity by emission of energy in the near to far red range.

In many applications it is desirable to derivatize compounds of general formula I to render them hydrophobic and permeable through cell membranes. The derivatizing groups should undergo hydrolysis inside cells to regenerate the compounds of general formula I and trap them inside the cells. For this purpose, it is preferred that any phenolic hydrox 's or free amines in the dye structures are acylated with  $C_1$ - $C_4$ acyl groups (e.g. formyl, acetyl, n-butryl) or converted to various other esters and carbonates [for examples, as described in Bundgaard, H., Design of Prodrugs, Elsevier Science Publishers (1985), Chapter I, page 3 et seq.]. Phenols can also be alkylated with 1-(acyloxy)alkyl, acylthiomethyl, acyloxy-alpha-benzyl, deltabutyrolactonyl, or methoxycarbonyloxymethyl groups. In the case of fluoresceins, rhodols, and rhodamines this manipulation is particularly useful, as it also results in conversion-of the acid moiety in these dyes to the spirolactone. To promote membrane permeation, the carboxyl at the 4-position of the cephalosporin should be esterified with 1-(acyloxy)alkyl, acylthiomethyl, acyloxy-alpha-benzyl, delta-butyrolactonyl, methoxycarbonyloxymethyl, phenyl, methylsulfinylmethyl, betamorpholionethyl, 2-(dimethylamino)ethyl, 2-(diethylamino) ethyl, or dialkylaminocarbonyloxymethyl groups as discussed in Ferres, H. (1980) Chem. Ind. 1980: 435-440. The most preferred esterifying group for the carboxyl is acetoxymethyl.

A general method for synthesis of compounds of general formula I is depicted below. As one of ordinary skill in the art will appreciate, the methods below can be used for a variety of derivatives, and other methods of synthesis are possible.

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In these compounds, RG is a nucleophile-reactive group (e.g., iodoacetamide, isocyanate, isothiocyanate, etc.); Nu is a nucleophile (e.g., -SH, -NH<sub>2</sub>, -OH, etc.); R<sub>0</sub> is H or an ester group (e.g., benzhydryl ester, tertiary butyl ester, etc.); Nu<sub>0</sub> is a bidentate nucleophile (e.g., HS<sup>-</sup>, HSCH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>, xanthate, etc.); and Hal is a halogen (e.g., chlorine, bromine or iodine).

The cephalosporin starting materials are commercially available cephalosporin derivatives 7-aminocephalosporanic acid or 7-amino 3'-chlorocephalosporanic acid as its benzhydryl or tertiary butyl ester (R<sub>0</sub>). Prior to coupling the dyes A and B carrying nucleophile reactive groups (RG) it is sometimes advantageous to esterify or alkylate their phenolic and free amine residues. The order of attaching dye A and dye B depends on the choice of reagents. Dye A is tethered to the cephalosporin via an alkyl amide linker. This is achieved by reacting a dye A carrying a nucleophile-reactive group (RG) with a bifunctional aliphatic acid (e.g., amino-, mercapto- or hydroxyalkyl acid) and coupling of the acid to the cephalosporin 7-amine (path 1). Alternatively, dye A carrying a nucleophilic group (e.g.,

W 96/30540 PCT/US96/04059

27

amine or thiol) is react d with a halogenat d alkyl acid and the acid c upled to the cephalosporin 7-amine (path 2). In both pathways, the order of the two reactions can be reversed. Dyes A containing an aliphatic acid can be directly coupled to the cephalosporin (path 3). Dye B carrying a nucleophilic substituent can be coupled to the 3'-position in the cephalosporin by direct displacement of the leaving group (LG) (path 4). A Dye B carrying a nucleophile-reactive group can be reacted with a bidentate nucleophile which is coupled then attached to the cephalosporin by leaving group (LG) displacement (path 5); the order of the reactions can be reversed.

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In some cases it might be necessary to conduct the first reaction with a bidentate nucleophile with one of its nucleophilic groups masked. The second coupling is then performed after removal of that protection group. After attachment of both dyes the cephalosporin ester is cleaved (in cases where R<sub>0</sub> is not H). To make membrane permeant substrates the acid-is then re-esterified to esters that can be deprotected by the cytoplasmic environment of a mammalian cell. For applications not involving cell cytoplasm, any remaining acyl and alkyl groups that were used to mask phenols and free amines on the dyes are removed.

preferred combinations of classes of donors and acceptors suitable for use in accordance with the present invention are indicated in Table 1. In embodiments of compounds of general formula I using these combinations, fluorescent resonant energy transfer (FRET) occurs. Of course, as would be readily understood by those working in the field, many other combinations of donors and acceptors/quenchers (including those that re-emit and those that do not) would be suitable for use in accordance with the present invention. In general, suitable donor and acceptor pairs are those where the donor's emission spectrum significantly overlaps the acceptor's excitation spectrum.

WO 96/30540 PCT/US96/04059

28

# TABLE 1 DONORS

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5	ACCEPTORS	II - VIII, XIX - XXI, XXIII - XXXIV	VII - XIV, XVII, XXII	XV - XVI, LV		
	II - VIII, XIX - XXI, XXIII - XXXIV	FRET				
0	VII - XIV, XVII, XXII	FRET	FRET	·		
	XV - XVII	FRET	FRET	FRET		
_	XL - XLV, XLVII - LII	FRET	FRET			
5	XXXV-XXXIX, XLVI-XLVII,	FRET	FRET	FRET		

Fluorescent donor moieties of particular interest
fluorescent donor moieties of particular quenchers of
include coumarins and fluoresceins. Particular quenchers of
interest include fluoresceins, rhodols and rhodamines.
Combinations of interest include the use of a coumarin donor
with a fluorescein, rhodol or rhodamine quencher, and a
fluorescein donor with a rhodol or rhodamine quencher.

Specific combinations of interest include the following: a
coumarin (e.g., 7-hydroxycoumarin) or chloro derivative
thereof with a fluorescein or dichloro derivative thereof; a

fluorescein with an eosin or tetrachlorofluorescein; a fluorescein with a rhodol derivative; and a rhodamine with a fluorescein.

Europium chelate donors may be suitably paired with

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Europium chelate donors may be suitably paired with acceptors of the formulae XV-XVII, XXXVI, XLVI-XLVII, LIV, and LVI. Terbium chelate donors may be suitably paired with acceptors of the formulae VIII-XVIII, XXXVI-XLI, and XLV-LIV, and LVI. The europium and terbium chelate donors may be of particular interest for their very narrow emission peaks and their microsecond-to-millisecond excited state lifetimes, which can be readily discriminated from background fluorescence and scattering with excited-state lifetimes of nanoseconds or less.

In many applications it is desirabl to derivatize compounds of general formula I to render them more hydr phobic and permeable through cell m mbranes. The derivatizing groups should undergo hydrolysis inside cells to regenerate the compounds of general formula I and trap them inside the cells. 5 For this purpose, it is preferred that any phenolic hydroxyls or free amines in the dye structures are acylated with C1-C4 acyl groups (e.g. formyl, acetyl, n-butyryl) or converted to various other esters and carbonates [for example, as described in Bundgaard, H., Design of Prodrugs, Elsevier Science D Publishers (1985), Chapter 1, page 3 et seq.]. Phenols can also be alkylated with 1-(acyloxy)alkyl, acylthiomethyl, acyloxy-alpha-benzyl, delta-butyrolactonyl, or methoxycarbonyloxymethyl groups. In the case of fluoresceins, rhodols and rhodamines, acylation or alkylation of the free 5 phenolic groups is particularly useful, as it also results in conversion of the acid moiety in these dyes to the spirolactone. To promote membrane permeation, the carboxyl at ---the 4-position of the-cephalosporin should be esterified with 1-(acyloxy)alkyl, acylthiomethyl, acyloxy-alpha-benzyl, delta-0 butyrolactonyl, methoxycarbonyloxymethyl, phenyl, methylsulfinylmethyl, beta-morpholinoethyl, 2-(dimethylamino)ethyl, 2-(diethylamino)ethyl, or dialkylaminocarbonyloxymethyl groups as discussed in Ferres, H. (1980) Chem. Ind. 1980: 435-440. The most preferred esterifying group for the carboxyl is acetoxymethyl.

The cephalosporin backbone serves as a cleavable
linker between two dyes. After cleavage it provides the
charges necessary to keep one of the two dyes inside the cell.

Dyes are chosen in a manner that one dye absorbs light
(quencher or acceptor chromophore) at the wavelength that the
other one emits (donor fluorophore). In the intact
cephalosporin the two dyes are in close proximity to each
other. When exciting the donor fluorophore one observes
fluorescence resonance energy transfer (FRET) from the donor
to the acceptor instead of donor fluorescence [Forster, T.,
Ann. Physik 2: 55-75 (1948)]. If the acceptor is a
nonfluorescent dye the energy is given off to the solvent; the

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donor fluorescence is quenched. In the case of the accept r being itself a fluorescent dye, fluorescenc re-emission occurs at the acceptor's emission wavelength. In polar solvents such as water, hydrophobic donor and acceptor fluorophores can stack when separated by a short flexible linker. Due to this association in the ground state, a \*dark complex" is formed [Yaron, A. et al., Anal. Biochem. 95: 228-235 (1979)]. In this complex, neither fluorophore can emit light, causing the fluorescence of both dyes to be quenched [Bojarski, C. and Sienicki, K. Energy transfer and migration in fluorescent solutions. In: Photochemistry and Photophysics, edited by Rabek, J.F. Boca Raton: CRC Press, Inc., 1990, pp. 1-57]. In either case, a large change in fluorescence goes along with  $\beta$ -lactam cleavage, which can be used to measure  $\beta$ -lactamase activity. As both dyes diffuse away from each other, stacking and energy transfer are disrupted. Cephalosporins carrying a donor and an acceptor dye which fluoresces are referred to herein as FRETcephalosporins. . ....

Fluorescence resonance energy transfer has been used as a spectroscopic ruler for measuring molecular distances in proteins and peptides as it is effective in the range from 10-100Å. This energy transfer is proportional to the inverse sixth power of the distance between donor and acceptor. Its efficiency is higher, the better donor emission and acceptor absorbance overlap, and the longer the fluorescence lifetime of the donor (in absence of the acceptor). FRET can be very efficient over distances of 10-20Å.

In the cephalosporin, distances for attachment of donor and acceptor are greater than 10Å and a minimum of 10 bond-lengths, if one includes the two minimal spacers at 7-and 3-positions. Over this distance FRET is very efficient, if the right donor-acceptor pairs are chosen. Conveniently, in a FRET-cephalosporin the 7-amine tethered dye stays attached to the polar hydrolysis products of cephalosporin cleavage, trapping it in the cells' cytoplasm. This position s best occupied by the donor fluorophore, although in some instances the acceptor may occupy this position. Upon

cleavage, fluorescence increases du to loss of the quencher dye.

The acceptor fluorophore is gen rally attached by a linker which imparts the greatest stability of the substrate to nucleophilic attack. A preferred linker is a thioether bond (-S-), which is very stable and due to its inductive effect reduces the reactivity of the  $\beta$ -lactam ring toward nucleophiles [Page, M.I., Adv.Phys.Org.Chem. 23: 165-270 (1987)]. In addition, the free thiol or thiolate group released upon hydrolysis often quenches the attached fluorophore, adding to the desired large change in fluorescence upon hydrolysis.

The fluorogenic substrates of the invention are initially colorless and nonfluorescent outside cells. The substrates are designed so they readily cross cell membranes into the cytoplasm, where they are converted to fluorescent compounds by endogenous nonspecific esterases and stay trapped due to their charges. In the intact molecules, fluorescence energy transfer occurs leading to fluorescence at a particular wavelength when the substrates are excited. Lactamase cleavage of the ß-lactam ring is followed by expulsion of the fluorescein molety with loss of fluorescence energy transfer. Excitation of the modified substrate now results in fluorescence at a different wavelength.

The assay systems of the present invention further provide an advantageous and rapid method of isolation and clonal selection of stably transfected cell lines containing reporter genes and having the desired properties which the transfection was intended to confer, e.g. fluorescent signal response after activation of a transfected receptor with a high signal-to-noise ratio from a high proportion of isolated cells. Current procedures for clonal selection of satisfactorily transfected, genetically engineered cells from the initial population, are done mainly by replica plating of colonies, testing of one set of colonies, visual selection of preferred clones, manual isolation of the replicas of the preferred clones by pipetting, and prolonged cellular cultivations. This procedure is laborious and time-consuming;

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it may require several m nths to generate a clone useful f r assays suited to drug screening. Moreover, it is difficult to manually sel ct and maintain more than a few hundred clones. Using the assays of this present invention, the desired signal from cellular beta-lactamase reporter system can be maintained within living and viable cells. Replica plating of colonies is unnecessary because single cells can be assayed and remain viable for further multiplication. Thus, from the population of initially transfected cells, one can rapidly select those few individual living cells with the best fluorescent signal, using automated instruments such as a fluorescent-activated cell sorter, e.g. the Becton Dickinson FACS VantageTM. The selected cells are then collected for cultivation and propagation to produce a clonal cell line with the desired properties for assays and drug screening.

working in the field, the combination of a novel substrate in accordance with the invention and a suitable ß-lactamase may be employed in a wide variety of different assay systems (such as are described in U.S. Patent 4,740,459). In particular, the fluorogenic substrates of the invention enable the detection of ß-lactamase activity in a wide variety of biologically important environments, such as human blood serum, the cytoplasm of cells and intracellular compartments; this facilitates the measurement of periplasmic or secreted ß-lactamase.

Further, the expression of any target protein can be detected by fusing a gene encoding the target protein to a  $\beta$ -lactamase gene, which can be localized by immunostaining and fluorescence or electron microscopy. For example,  $\beta$ -lactamase fusion proteins may be detected in the lumen of organelles through the use of the substrates of the invention; only subcellular compartments containing the fusion protein fluoresce at a wavelength characteristic of the cleaved substrate, whereas all others fluoresce at a wavelength characteristic of the intact molecule.

Both the intact and cleaved substrate are well retained in cells without the use of special measures, such as

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chilling. Th color change (even in individual small mammalian cells) is visible through a fluoresc nce microscope using normal color vision or photographic film; the fluorescence signal may be quantified and further enhanced by conventional digital image processing techniques. Moreover, because gene activation is detected not by a change in a single intensity but rather by a color change or a change in the ratio between two intensities at different wavelengths, the assays of the present invention are relatively immune to many artifacts such as variable leakiness of cells, quantity of substrate, illumination intensity, absolute sensitivity of detection and bleaching of the dyes.

A variety of substrates (e.g., compounds of general formulas 17, 22 and 25) have been prepared and their emission spectra obtained before and after ß-lactamase cleavage. These substrates allow for ß-lactamase detection primarily in vitro, as they bind strongly to serum and cellular proteins. Due to their hydrophobic nature, the fluorophores stack; this leads to a loss of fluorescence in the intact substrate. ß-lactamase cleaves the substrates and relieves the stacking, allowing for fluorescence. Compounds (e.g., compound 11, Example 1) with reversed location of donor and acceptor fluorophore on the cephalosporin exhibit similar fluorescence behavior.

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In one preferred embodiment of the invention, a compound of general formula 1 was coupled to a compound of general formula 2 to form a compound of general formula 3. Commercially-available compound 4 was then coupled to compound 3 using dicyclohexylcarbodiimide and the product reacted with compound 5, yielding a compound of general formula 6. Deprotection of compound 6 generated a compound of general formula 7. In exemplary embodiments, Acyl was acetyl, R\* was Me and R\* H (a), or Acyl was butyryl, R\* was H and R\* Cl (b); R\* was trimethylsilyl or benzyl.

The compounds of gen ral formula 6 were modified t obtain membrane permeant derivatives which were c nverted t the corresponding flu resc nt compounds of general f rmula 7 in intact cells due to the action of ndogenous nonspecific esterases. In these molecules, fluorescence resonance energy transfer occurs from the 7-hydroxycoumarin moiety to the fluorescein moiety, leading to green fluorescence when the compounds are excited at about 400 nm. After cleavage of the 8-lactam ring, excitation of the 7-hydroxycoumarin moiety results in blue fluorescence; in exemplary embodiments, a 25fold increase in fluorescence at about 450 nm and a three- to fourfold decrease in fluorescence at 515 nm was observed.

# MONITORING GENE EXPRESSION

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The substrates of this invention make it feasible to use  $\beta$ -lactamase as a reporter gene to monitor the expression from a set of expression control sequences. In one aspect, this invention provides methods for monitoring gene expression from a set of expression control sequences by using  $\beta$ lactamase as a reporter gene. A cell is provided that has been transfected with a recombinant nucleic acid molecule. comprising the expression control sequences operably linked to nucleic acid sequences coding for the expression of  $\beta$ lactamase.

## Recombinant Nucleic Acids

As used herein, the term "nucleic acid molecule" includes both DNA and RNA molecules. It will be understood that when a nucleic acid molecule is said to have a DNA sequence, this also includes RNA molecules having the corresponding RNA sequence in which "U" replaces "T." term "recombinant nucleic acid molecule" refers to a nucleic acid molecule which is not naturally occurring, and which comprises two nucleotide sequences which are not naturally joined together. Recombinant nucleic acid molecules are produced by artificial combination, e.g., genetic engineering techniques or chemical synthesis.

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Nucleic acids enc ding  $\beta$ -lactamases can be brained by methods known in the art, for example, by polymerase chain reaction of cDNA using primers based on the DNA sequence in Fig. 1. PCR methods are described in, for example, U.S. Pat. No. 4,683,195; Mullis et al. (1987) Cold Spring Harbor Symp. Quant. Biol. 51:263; and Erlich, ed., PCR Technology, (Stockton Press, NY, 1989).

The construction of expression vectors and the expression of genes in transfected cells involves the use of molecular cloning techniques also well known in the art.

Sambrook et al., Molecular Cloning -- A Laboratory Manual, Cold Spring Harbor Laboratory, Cold Spring Harbor, NY, (1989) and Current Protocols in Molecular Biology, F.M. Ausubel et al., eds., (Current Protocols, a joint venture between Greene Publishing Associates, Inc. and John Wiley & Sons, Inc., (most recent Supplement)).

Nucleic acids used to transfect cells with sequences coding for expression of the polypeptide of interest generally will be in the form of an expression vector including expression control sequences operatively linked to a nucleotide sequence coding for expression of the polypeptide. As used, the term nucleotide sequence "coding for expression of\* a polypeptide refers to a sequence that, upon transcription and translation of mRNA, produces the polypeptide. As any person skilled in the are recognizes, this includes all degenerate nucleic acid sequences encoding the same amino acid sequence. This can include sequences containing, e.g., introns. As used herein, the term "expression control sequences" refers to nucleic acid sequences that regulate the expression of a nucleic acid sequence to which it is operatively linked. Expression control sequences are "operatively linked" to a nucleic acid sequence when the expression control sequences control and regulate the transcription and, as appropriate, translation of the nucleic acid sequence. Thus, expression control sequences can include appropriate promoters, enhancers, transcription terminators, a start codon (i.e., ATG) in front of a proteinencoding gene, splicing signals for introns, maintenance of

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the correct reading frame of that gen t permit proper translation of the mRNA, and stop codons.

The recombinant nucleic acid can be incorporated into an expression vector comprising expression control sequences operatively linked to the recombinant nucleic acid. The expression vector can be adapted for function in prokaryotes or eukaryotes by inclusion of appropriate promoters, replication sequences, markers, etc.

The recombinant nucleic acid used to transfect the cell contains expression control sequences operably linked to a nucleotide sequence encoding a  $\beta$ -lactamase. The  $\beta$ -lactamase encoded can be any known to the art or described herein. This includes, for example, the enzymes shown in Fig. 7.

This invention provides novel recombinant nucleic acid molecules including expression control sequences adapted for function in a non-mammalian eukaryotic cell operably linked to a nucleotide sequence coding for the expression of a cytosolic  $\beta$ -lactamase. As used herein, "cytosolic  $\beta$ -lactamase" refers to a  $\beta$ -lactamase that lacks amino acid sequences for secretion from the cell membrane, e.g., the signal sequence. For example, in the polypeptide of Sequence 1 of Fig. 7, the signal sequence has been replaced with the amino acids Met-Ser. Accordingly, upon expression, this  $\beta$ -lactamase remains within the cell.

This invention provides recombinant nucleic acid molecules including expression control sequences adapted for function in a mammalian eukaryotic cell operably linked to a nucleotide sequence coding for the expression of a  $\beta$ -lactamase.

It is further preferable that the ribosome binding site and nucleotide sequence coding for expression of  $\beta$ -lactamase contain sequences preferred by mammalian cells. Such sequences improve expression of  $\beta$ -lactamase in mammalian cells. Preferred sequences for expression in mammalian cells are described in, for example, Kozak, M., J. Cell Biol. 108: 229-241 (1989), referred to herein as "Kozak sequences". The nucleotide sequence for cytosolic  $\beta$ -lactamase in Sequence 3 of

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Fig. 7 contains Kozak sequences for the nucleotides -9 t 4 (GGTACCACCATGA).

When used in mammalian cells, the expressi n c ntrol sequences are adapt d for function in mammalian cells. method of this invention is useful to testing expression from any desired set of expression control sequences. particular, this invention is useful for testing expression from inducible expression control sequences. As used herein, \*inducible expression control sequences\* refers to expression control sequences which respond to biochemical signals either by increasing or decreasing the expression of sequences to which they are operably linked. For example, in the case of genes induced by steroid hormones, the expression control sequences includes hormone response elements. The binding of a steroid hormone receptor to the response element induces transcription of the gene operably linked to these expression control sequences. Expression control sequences for many genes and for inducible genes, in particular, have been isolated and are well known in the art. The invention also is useful with constitutively active expression control sequences.

The transfected cell is incubated under conditions to be tested for expression of  $\beta$ -lactamase from the expression control sequences. The cell or an extract of the cell is contacted with a  $\beta$ -lactamase substrate of the invention under selected test conditions and for a period of time to allow catalysis of the substrate by any  $\beta$ -lactamase expressed. Then the donor moiety from this sample is excited with appropriate ultraviolet or visible wavelengths. The degree of fluorescence resonance energy transfer in the sample is

If the cell did not express  $\beta$ -lactamase, very little of the substrate will have been cleaved, the efficiency of FRET in the cell will be high, and the fluorescence characteristics of the cell or sample from it will reflect this efficiency. If the cell expressed a large amount of  $\beta$ -lactamase, most of the substrate will be cleaved. In this case, the efficiency of FRET is low, reflecting a large amount

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or high efficiency of the cleavage enzyme relative to the rate of synthesis of the tandem flu rescent protein construct. In one aspect, this method can be used to compare mutant cells to identify which ones possess greater or less enzymatic activity. Such cells can be sorted by a fluorescent cell sorter based on fluorescence.

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Also, as will be apparent to those working in the field of using reporter gene cell-based assays for screening samples or pools of samples (such as compounds (combinatorial or synthetic), natural product extracts, or marine animal extracts) to identify potential drug candidates which act as agonists, inverse agonists or antagonists of cellular signaling or activation, the combination of cells (preferably mammalian) genetically engineered to express beta-lactamase under the control of different regulatory elements/promoters and the use of the novel beta-lactamase substrate compounds of the present invention will provide distinct advantages over known reporter genes (including, but not limited to, chloramphenicol acetyl transferase, firefly luciferase, bacterial luciferase, Vargula luciferase, aequorin, betagalactosidase, alkaline phosphatase) and their requisite substrates.

By the choice of appropriate regulatory elements and promoters to control expression of beta-lactamase, assays can be constructed to detect or measure the ability of test substances to evoke or inhibit functional responses of intracellular hormone receptors. These include expression control sequences responsive to inducible by mineralcorticosteroids, including dexamethasone [J. Steroid Biochem. Molec. Biol. Vol. 49, No. 1 1994, pp.31-3]), gluococorticoid, and thyroid hormone receptors [as described in US patent 5,071,773]. Additional such intracellular receptors include retinoids, vitamin D3 and vitamin A [Leukemia vol 8, Suppl. 3, 1994 ppS1-S10; Nature Vol. 374, 1995, p.118-119; Seminars in Cell Biol., Vol. 5, 1994, p.95-103]. Specificity would be enabled by use of the appropriate promoter/enhancer element. Additionally, by choice of other regulatory elements or specific promoters, drugs which

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influ nc expression of specific genes can be identified. Such drugs could act n specific signaling molecules such as kinases, transcription factors, or molecules such signal transducers and activators of transcription [Science Vol. 264, 1994, p.1415-1421; Mol. Cell Biol., Vol. 16, 1996, p.369-375). Specific microbial or viral promoters which are potential drug targets can also be assayed in such test systems.

Also by the choice of promoters such as c-fos or cjun [US patent 5,436,128; Proc. Natl. Acad. Sci. Vol. 88, 1991, pp. 5665-5669] or promoter constructs containing regulatory elements responsive to second messengers [Oncogene, 6: 745-751 (1991)] (including cyclic AMP-responsive elements, phorbol ester response element (responsive to protein kinase C activation), serum response element (responsive to protein kinase C-dependent and independent pathways) and Nuclear Factor of Activated T-cells response element (responsive to calcium) to control expression of beta-lactamase, assays can be constructed to detect or measure substances or mixtures of substances that modulate cell-surface receptors including, but not limited to, the following classes: receptors of the cytokine superfamily such as erthyropoietin, growth hormone, interferons, and interleukins (other than IL-8) and colonystimulating factors; G-protein coupled receptors [US patent 5,436,128] for hormones; such as calcitonin, epinephrine or gastrin, pancrine or autocrine mediators, such as stomatostatin or prostaglandins, and neurotransmitters such as norepinephrine, dopamine, serotonin or acetylcholine; tyrosine kinase receptors such as insulin growth factor, nerve growth factor [US patent 5,436,128]. Furthermore, assays can be constructed to identify substances that modulate the activity of voltage-gated or ligand-gated ion channels, modulation of which alters the cellular concentration of second messengers, particularly calcium [US patent 5,436,128]. Assays can be constructed using cells that intrinsically express the promoter, receptor or ion channel of interest or into which 35 the appropriate protein has been genetically engineered.

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The expression contr 1 sequences also can be those responsiv to substances that modulate cell-surface receptors or that modulate intra-cellular receptors.

To measure whether a substance or mixture of substances activates extracellular or intracellular receptors or other cellular responses, cells containing beta-lactamase controlled by a desired promoter/enhancer element are incubated with test substance(s), substrate then added, and after a certain period of time the fluorescence signal is measured at either one or two excitation-emission pairs appropriate to the chosen compound of the invention (e.g. compound CCF2 with wavelength pairs of near 405 nm and near 450 nm and near 405 and near 510 nm). This fluorescent result is compared to control samples which have had no drug treatment and, when feasible, control samples with a known inhibitor and a known activator. The effect of any active drugs is then determined using the ratio of the fluorescence signal found in test wells to the signals found in wells with - no drug treatment. Assays are performed in wells in a microtiter plate containing 96 or more wells or in an assay system with no compartments such as a gel matrix or moist membrane environment. Detection could be done for example by microtiter plate fluorimeters, e.g. Millipore Cytofluor, or imaging devices capable of analyzing one or more wells or one or more assay points in a certain surface area, e.g. as supplied by Astromed. The ability to retain the substrate in the cytoplasm of living cells is advantageous as it can allow a reduction in signal interference from coloured or quenching substances in the assay medium. Furthermore, the fluorescent signal from the compounds of this invention, such as CCF2, can be readily detected in single cells and thus allowing assay miniaturization and an increased number of tests per surface Miniaturized assays also further increase the throughput of an imaging detection system as there are more samples within the imaging field.

The assay systems of the present invention further provide an advantageous and rapid method of isolation and clonal selection of stably transfected cell lines containing

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reporter genes and having th desir d properti s which the transfection was intended to confer, e.g. fluorescent signal response after activation of a transfected receptor with a high signal-to-noise ratio of at least 10:1 from a high proportion of isolated cells. Current procedures for clonal selection of satisfactorily transfected, genetically engineered cells from the population initial transfected with the vectors of interest, are done mainly by manual means and involve several rounds of microscopic analyses, selecting the visually preferred clone, isolation of the clone by manual pipetting stages and prolonged cellular cultivations. This procedure is laborious and time-consuming; it may require several months to generate a clone useful for assays suited to drug screening. Moreover, it is difficult to manually select and maintain more than a few hundred clones. Using the assays of this present invention, the desired signal from cellular beta-lactamase reporter system can be maintained within living and viable cells. Thus, one can rapidly select, from the population of initially transfected cells, those few living cells with the best fluorescent signal using automated instruments such as a fluorescent-activated cell sorter, e.g. the Becton Dickinson FACS Vantage. The selected cells are then collected for cultivation and propagation to produce a clonal cell line with the desired properties for assays and drug screening.

In addition, the presence (for example, in human serum, pus or urine) of bacteria resistant to ß-lactam antibiotics can be readily detected using the substrates of the present invention. Only in the presence of an active ß-lactamase is there a change in the fluorescence spectrum from that of the intact molecule to one characteristic of the cleavage product. The substrates of the present invention are superior to prior art chromogenic substrates Nitrocephin and PADAC, in that the inventive substrates are stable to human serum. The novel substrates are also more sensitive than the chromogenic substrate CENTA, because they experience a much smaller optical background signal from human serum and a lower detection limit for fluorescence versus absorbance.

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The invention may be better understood with reference to the accompanying examples, which are intended for purposes of illustration only and should not be construed as in any sense limiting the scope of the invention as defined in the claims appended hereto.

#### **MEASUREMENTS**

The degree of FRET can be determined by any spectral or fluorescence lifetime characteristic of the excited construct, for example, by determining the intensity of the fluorescent signal from the donor, the intensity of fluorescent signal from the acceptor, the ratio of the fluorescence amplitudes near the acceptor's emission maxima to the fluorescence amplitudes near the donor's emission maximum, or the excited state lifetime of the donor. For example, cleavage of the linker increases the intensity of fluorescence from the donor, decreases the intensity of fluorescence from the acceptor, decreases the ratio of fluorescence amplitudes from the acceptor to that from the donor, and increases the excited state lifetime of the donor.

preferably, changes in the degree of FRET are determined as a function of the change in the ratio of the amount of fluorescence from the donor and acceptor moieties, a process referred to as "ratioing." Changes in the absolute amount of substrate, excitation intensity, and turbidity or other background absorbances in the sample at the excitation wavelength affect the intensities of fluorescence from both the donor and acceptor approximately in parallel. Therefore the ratio of the two emission intensities is a more robust and preferred measure of cleavage than either intensity alone.

The excitation state lifetime of the donor moiety is, likewise, independent of the absolute amount of substrate, excitation intensity, or turbidity or other background absorbances. Its measurement requires equipment with nanosecond time resolution, except in the special case of lanthanide complexes in which case microsecond to millisecond resolution is sufficient.

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Additional suitable Chel moieties are described in Vallarino, L.M., & Leif, R.C., US Pat. 5,373,093; Sabbatini, N. et al, Pure and Applied Chem. 67: 135-140 (1995); Mathis, G., Clinical Chem. 41: 1391-1397 (1995); Horiguchi, D., Chem. Pharm. Bull. 42: 972-975 (1994); Takalo, H. et al, Bioconjugate Chem. 5: 278-282 (1994); Saha, A.K. et al, J. Amer. Chem. Soc. 115: 11032 (1993); Li, M. & Selvin, P.R., J. Amer. Chem. Soc. 117: 8132-8138 (1995).

Fluorescence in a sample is measured using a In general, excitation radiation, from an fluorimeter. excitation source having a first wavelength, passes through excitation optics. The excitation optics cause the excitation radiation to excite the sample. In response, fluorescent proteins in the sample emit radiation which has a wavelength that is different from the excitation wavelength. Collection optics then collect the emission from the sample. The device can include a temperature controller to maintain the sample at a specific temperature while it is being scanned. According to one embodiment, a multi-axis translation stage moves a microtiter plate holding a plurality of samples in order to position different wells to be exposed. The multi-axis translation stage, temperature controller, auto-focusing feature, and electronics associated with imaging and data collection can be managed by an appropriately programmed digital computer. The computer also can transform the data collected during the assay into another format for presentation.

Methods of performing assays on fluorescent materials are well known in the art and are described in, e.g., Lakowicz, J.R., Principles of Fluorescence Spectroscopy, New York:Plenum Press (1983); Herman, B., Resonance energy transfer microscopy, in: Fluorescence Microscopy of Living Cells in Culture, Part B, Methods in Cell Biology, vol. 30, ed. Taylor, D.L. & Wang, Y.-L., San Diego: Academic Press (1989), pp. 219-243; Turro, N.J., Modern Molecular Photochemistry, Menlo Park: Benjamin/Cummings Publishing Col, Inc. (1978), pp. 296-361.

#### **EXAMPLES**

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All silica gel chromatography was perform d using silica gel (Merck, grade 60, 230-400 mesh, 60Å) purchased from Aldrich. Bakerbond Octadecyl from J. T. Baker was used for C<sub>10</sub> reverse phase chromatography. Solvents (high pressure liquid chromatography grade) were used for chromatography as received, or dried over activated molecular sieves (3 Å) for synthetic purposes.

Fluorescence excitation and emission spectra were measured either on a Spex Fluorolog 111 or on a K2 fluorometer (ISS, Champaigne, IL) in ratio mode with a rhodamine B quantum counter. The efficiency of fluorescence energy transfer was determined from the change in the integrated fluorescence emission at the donor emission wavelength upon treatment with B-lactamase. For fluorescence microscopy imaging, two different imaging setups were used. One, with an inverted fluorescence microscope, Zeiss IM-35 (Thornwood, NY) coupled to a silicon-intensified target (SIT) camera (Dage-MTI, Michigan City, IN) has been described in detail [Tsien, R.Y. (1986) New tetracarboxylate chelators for fluorescence measurement and photochemical manipulation of cytosolic free calcium concentrations, in: Optical Methods in Cell Physiology, ed.-de Weer, P. & Salzberg, B., New York: Wiley, pp. 327-345; Tsien and Harootunian (1990) Cell Calcium 11:93-109]. The other consisted of a cooled charge-coupled-device (CCD) camera (Photometrics, Tucson, AZ) connected to an inverted fluorescence microscope (Zeiss Axiovert).

Fluorescence resonance energy transfer was measured by monitoring the ratio of fluorescence intensities at donor and acceptor emission wavelengths using commercially-available filters (Omega Optical).

Excitation: 360 DF 40, dichroic mirror 390 DCLP

or 405 DF 15, dichroic mirror 420 DRLPO2

Emission: 450 DF 65 (donor emission)

515 EFLP (acceptor emission)

435 EFLP (to view donor and acceptor

fluorescence simultaneously)

# Example 1 (Compound 11)

To test the optical properties of a cephalosporin with two dye molecules attached th following model compound was synthesized.

The first synthesis step was to convert 7-aminocephalosporanic acid into a bifunctional cephalosporin carrying a thiol in the 0 3'-position and the 7-amine [Van Heyningen, E. and Brown, C.N., J.Med.Chem. 8: 174-181 (1965); Japanese Patent, Kokai 75/18494, CA 85, 97320d]. This cephalosporin was then reacted selectively with an thiol-reactive dye, followed by an aminereactive dye. The thiol-reactive dye 5, (6) -iodoacetamido-:5 fluorescein and the amine-reactive dye 5, (6)-carboxy-N,N,N',N'-tetramethylrhodamine-succinimide were coupled to the cephalosporin in aqueous dimethylformamide at pH 8. The product will be referred to as RCF. 30

In phosphate buffer at pH 7 RCF is virtually non fluorescent; neither fluorescein nor rhodamine show much fluorescence when excited at their respective excitation maxima, which is indicative of chromophore stacking ("dark complex"). After long term treatment with  $\beta$ -lactamase the  $\beta$ -lactam is cleaved causing the fluorescence of both dyes to reappear (Figs. 1(a) and 1(b)). This experiment confirms that one can measure  $\beta$ -lactamase catalyzed hydrolysis of the SUBSTITUTE SHEET (RULE 26)

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 $\beta$ -lactam in cephalosporins by the loss of flu rescence quenching using an appropriate donor-acceptor pair.

#### Example 2

The thiomethyl linker was introduced by conversion of 5-fluoresceinamine to 5-mercaptofluorescein via diazotization, conversion to the ethylxanthate, and degradation of the xanthate by aqueous acid to the free sulfhydryl. It was coupled to 7-bromoacetamido-cephalosporanic acid by nucleophilic displacement of the bromide by the mercapto group of the fluorescein.
7-Bromoacetamido-cephalosporanic acid had been prepared from 7-aminocephalosporanic acid and bromoacetyl bromide [Bunnell, C.A. et al. Industrial manufacture of cephalosporins. In:
Beta-Lactam Antibiotics for Clinical Use. Series: Clinical Pharmacology Vol. 4, edited by Queener, S.F., Webber, J.A. and Queener, S.W. New York: M. Dekker, 1986, p. 255-283].

To prepare

7β-[(5-diacetylfluorescein)thio]acetamido-3-(acetoxymethyl)-3-cephem-4-carboxylic acid (14), in a nitrogen atmosphere
130mg (0.29mmol) 5-fluoresceinthiol diacetate were dissolved in 10ml dimethylformamide and added to 120mg (0.31mmol)
7β-bromoacetamido-3-(acetoxymethyl)-3-cephem-4-carboxylic acid

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in 10ml 1M potassium ph sphat buffer adjusted to pH 8.0. solution was stirred for 8 hours at room temperature after which th solvents were removed in vacuo. The residue was dissolved in 10ml water and the pH of the solution was carefully adjusted to pH 5 with dilute phosphoric acid. At this point nonpolar byproducts precipitated and were removed by centrifugation. Further acidification to pH 2.7 precipitated the title compound which was collected by centrifugation, washed 3 times with 2 ml diethylethertetrachloromethane (1:2), and dried in vacuo. 'H NMR (CDCl<sub>3</sub>): δ 2.08ppm (s, 3H, acetate), δ 3.36ppm, 3.53ppm (2d, 2H, J=17.3 Hz, C-2), & 3.87ppm (s, 2H, side chain methylene), & 4.88ppm, 5.16ppm (2d, 2H, J=13.6 Hz, C-3'), 8 4.96ppm (d, 1H, J=4.9 Hz, C-6),  $\delta$  5.81ppm (dd, 1H, J<sub>1</sub>=8.2 Hz, J<sub>2</sub>=4.9 Hz, C-7),  $\delta$  6.85ppm (m, 4H, xanthene),  $\delta$  7.10 (s, 2H, xanthene),  $\delta$  7.15ppm (d, 1H, J=8.2 Hz, amide),  $\delta$  7.69ppm (d, 1H, J=8.2 Hz, phthalic),  $\delta$ 7.91ppm (d, 1H, J=8.2 Hz, phthalic), 8 8.11ppm (s, 1H, phthalic).

5-Fluoresceinamine was brominated to generate
5-eosinamine, which was converted into 5-mercaptoeosin in
analogous way to the 5-mercaptofluorescein. In a nucleophilic
displacement of the cephalosporin acetate by 5-mercaptoeosin
diacetate the FRET-cephalosporin was generated as the
protected tetraacetyl derivative.

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To prepare 5-eosinamine, 1.74g (5mmol) 5-fluoresceinamine was suspended in 30ml glacial acetic and 2.06ml (40mmol, 100% excess) bromine was added. With the addition of bromin the fluoresceinamine went into solution. The solution was heated for six hours at 90°C, during which period a white precipitate began to form. An ice-cooled trap attached to the flask kept bromine from escaping into the atmosphere. Excess bromine was then recovered by distillation into a liquid nitrogen cooled collecting flask. One volume of water was added to the acetic acid solution to precipitate any product remaining in solution. The precipitate was collected by filtration and dissolved in 1N aqueous sodium hydroxide. 5-Eosinamine was precipitated as the free amine by addition of glacial acetic acid. The eosinamine was dissolved in little chloroform and methanol was added. Concentrating this solution on the rotary evaporator gave 2.56g (3.85mmol, 77%) eosinamine as a fine white powder (the eosinaminespirolactone).

To prepare 5-eosin-ethylxanthate diacetate, 670mg (1mmol) 5-eosinamine were stirred in 2ml concentrated sulfuric acid and 2ml glacial acetic acid. The suspension was cooled with an ice-salt bath to a few degrees below 0°C, which turned it into a thick paste that was difficult to stir. (2.9mmol) sodium nitrite in 1ml water were added dropwise over the period of one hour. After another 2 hours at 0°C 20g of ice was slowly added. The flask was put on the high vacuum pump in the cold, to remove excess nitrous gases (caution!!). Saturated ice-cold aqueous sodium bicarbonate solution was added until the solids dissolved into the dark red solution. 200mg (1.2mmol) Potassium ethylxanthate was added and a pink precipitate formed (5-eosindiazonium xanthate). A few crystals of nickel(II)chloride catalyzed the conversion of the diazonium salt with evolution of nitrogen. Once nitrogen evolution had ceased the products were precipitated with 1N hydrochloric acid. The precipitate was collected by filtration and dried in vacuo. It was treated with acetic anhydride-pyridine (1:1) at 40°C for one hour. After removal of the reagents in vacuo, the residue was chromatographed over

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silica gel with ethyl acetate-hexane (1:4) as eluent. desired product eluted first. The yield was 110mg (0.13mmol, 13%) of the title compound as a white powder.

For preparation of the disulfid dimer of 5-eosinthiol diacetate (dimer of 15), 110mg (0.13mmol) 5-eosin-ethylxanthate diacetate was stirred in 10ml concentrated (30%) aqueous ammonia and the solution was heated to 70°C. Air was bubbled slowly through the solution to oxidize the thiol to the disulfide in situ. After 2 hours the solvents were removed on the rotary evaporator at 40°C and the residue was treated with acetic anhydride-pyridine (1:1). After removal of the reagents in vacuo the residue was chromatographed over silica gel with ethyl acetate-hexane (1:4) as the eluent. Yield was 90mg (60μmol, 91%) of the title product as a white powder. The compound was reduced to the monomer (15) by dissolving it in methanol with addition of sodium acetate and addition of 20 equivalents mercaptoethanol. After 2 hours the methanolic solution was poured into 3 volumes 5% aqueous acetic acid from which the precipitating 5-fluoresceinthiol monomer was collected by centrifugation. 20 The solid was washed with water until no odor of mercaptoethanol remained.

78-[(5-diacetylfluorescein)thio]acetamido-3-(acetoxymethyl)-3-cephem-4-carboxylic acid (14) and deacylation with acetylesterase was effected as follows. 10mg (13µmol) 78-[(5-Diacetylfluorescein)thio]acetamido-3-(acetoxymethyl)-3-cephem-4-carboxylic acid and 10mg (13 $\mu$ mol) diacetyl-5-eosinthiol were dissolved in 200µl dry acetonitrile and the solution was sealed under argon in a glass tube. The tube was kept in an oil bath at 84°C (± 2°C) for 16 hours. Then it was cut open, the solution transferred to a flask and the solvent removed in vacuo. The residue was flash-chromatographed over silica gel with ethyl acetate-methanol-acetic acid (100:1:1) as the eluent. Deprotection of the acetates was achieved by incubating the product with orange peel acetylesterase in 50mM phosphate buffer (pH7) for 24 hours at 37°C. The deacylated product was purified by C10 reverse phase chromatography. The

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eluent was a step gradient of 25mM aque us phosphate buff r (pH7) and methan 1. Fluorescein bypr ducts eluted with 33% and 50% methanol in the eluent, after which the desired product eluted in 66% methanol.

The deprotected compound shows little fluorescence in phosphate buffer as the two hydrophobic dyes stack. The remaining fluorescence is due to fluorescence resonance energy transfer (FRET). This compound is a good substrate for RTEM  $\beta$ -lactamase and will be referred to as PCE.

Cleavage of the compound increases fluorescence at 515nm about 70-fold (Fig. 2). The fluorescence properties of the compound can be attributed to dye-dimer formation, as FRET increases drastically once methanol is added to the solution. Methanol breaks the hydrophobic interaction that causes the fluorophores to stack.

## Example 3 (compound 22)

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The 3'-acetate f 7-aminocephalosporanic acid was displaced by ethylxanthate [Van Heyningen and Brown (1965), supra] which was hydrolysed to the free sulfhydryl with aqueous acetylhydrazine [Japanese Patent, Kokai 75/18494, CA85, 97320d]. The sulfhydryl group was reacted with 5-bromoacetamido-rhodol-X in aqueous dimethylformamide. The cephalosporin 7-amine was reacted with bromoacetyl bromide in aqueous dioxane, followed by bromide displacement with 5-fluoresceinthiol to yield a FRET-cephalosporin that is virtually nonfluorescent in 50mM phosphate buffer pH 7. This compound is referred to as FCRX.

The first step in preparation of 5-rhodol-Xbromoacetamide was synthesis of 9-(2'-carboxy-4'(5')-nitro-benzoyl)-8-hydroxyjulolidine and separation of 10.1g (48mmol,92% purity) 4-Nitrophthalic the isomers. anhydride were dissolved in 20ml toluene at 70°C. 9.76g (50mmol, 97% purity) 8-Hydroxyjulolidine in 20ml ethyl acetate were added and the solution kept at 70°C for 30min. The reaction mixture was run through a short bed of silica gel followed by ethyl acetate as eluent. The solvents were removed in vacuo and the solid redissolved in a minimum amount of refluxing ethyl acetate. The isomer with the nitro-group meta to the benzoic acid-crystallizes over night from solution in orange crystals (3.47g in first fraction). After additional fractional crystallization the pure isomer was <sup>1</sup>H NMR (CDCl<sub>3</sub>) of crystallized isomer: 8 1.91ppm (m, obtained. 4H, aliphatic methylenes), & 2.73ppm, 2.46ppm (2m, 4H, anilinic methylenes),  $\delta$  3.26ppm (m, 4H, benzylic methylenes),  $\delta$  6.32ppm (s, 1H, julolidine),  $\delta$  7.53ppm (d, 1H, J=8.4 Hz, phthalic),  $\delta$  8.43ppm (dd,  $J_1$ =8.4 Hz,  $J_2$ =2.2 Hz, phthalic),  $\delta$ 8.90ppm (d, 1H, J=2.2 Hz, phthalic).

For preparation of 5-rhodol-X-amine hydrochloride (named by analogy with rhodamine-X), 1.91g (5.0mmol)
9-(2'-Carboxy-4'-nitro-benzoyl)-8-hydroxyjulolidine was stirred in 5ml concentrated (96%) sulfuric acid. 700mg (6.4mmol, 1.25equ.) resorcinol was added with cooling over a period of 15 minutes. The suspension was stirred 1.5 hours at room temperature and then poured into 200ml water with

vigorous stirring. The purple precipitate was collected by filtration and rediss lved in 75ml water with the help of 5.3g (22mmol) sodium sulfide nonahydrate. 2.5g (44.6 mmol) Anhydrous sodium bisulfide was added and the solution refluxed for 24 hours. Then, after cooling to room temperature, the 5 product was precipitated by addition of glacial acetic acid. The solid was collected by filtration and boiled with 100ml half-saturated aqueous hydrochloric acid. The solution was filtered hot through a glass frit to remove sulfur. solution volume was reduced to 10ml on the rotary evaporator. 0 1 Volume saturated brine was added and the precipitate collected by filtration. Crystallization from refluxing hydrochloric acid yielded 1.78g (3.85mmol, 77%) dark red crystals of 5-rhodol-X-amine hydrochloride. 'H NMR (dDMSO) of 5-nitro-rhodol-X: & 1.90ppm, 2.05ppm (2m, 4H, aliphatic 5 methylenes),  $\delta$  2.72ppm, 3.03ppm (2m, 4H, anilinic methylenes),  $\delta$  3.66ppm (m, 4H, benzylic methylenes),  $\delta$  6.90ppm (s, 1H, xanthene),  $\delta$  6.96ppm (dd, 1H, J<sub>1</sub>=9.0 Hz, J<sub>2</sub>=2.1 Hz, xanthene), δ 7.11ppm (d, 1H, J=9.0 Hz, xanthene), δ 7.22ppm (d, 1H, J=2.1 Hz, xanthene),  $\delta$  7.78ppm (d, 1H, J=8.4 Hz, phthalic),  $\delta$ 8.70ppm (dd, 1H,  $J_1=8.4$  Hz,  $J_2=2.4$  Hz, phthalic),  $\delta$  8.91ppm (d, 1H, J=2.4 Hz, phthalic). 'H NMR (CD3OD) of 5-rhodol-X-amine hydrochloride: δ 2.00ppm, 2.14ppm (2m, 4H, aliphatic methylenes),  $\delta$  2.75ppm, 3.11ppm (2m, 4H, anilinic methylenes),  $\delta$  3.67ppm (m, 4H, benzylic methylenes),  $\delta$  6.85ppm (s, 1H, 5 xanthene),  $\delta$  6.94ppm (dd, 1H,  $J_1=9.0$  Hz,  $J_2=2.1$  Hz, xanthene),  $\delta$  7.13ppm (d, 1H, J=9.0 Hz, xanthene),  $\delta$  7.16ppm (d, 1H, J=2.1 Hz, xanthene),  $\delta$  7.55ppm (d, 1H, J=8.1 Hz, phthalic),  $\delta$ 7.82ppm (dd, 1H,  $J_1=8.1$  Hz,  $J_2=1.9$  Hz, phthalic),  $\delta$  8.28ppm (d, 1H, J=1.9 Hz, phthalic). 0

Preparation of 5-rhodol-X-bromoacetamide (18) was effected as follows. 115mg (0.25mmol) 5-Rhodol-X-amine hydrochloride were dissolved with 180mg (2.1mmol) sodium bicarbonate in 2ml water-dioxane (1:1). The solution was cooled on ice and  $175\mu l$  (2mmol) bromoacetylbromide were added with stirring over a period of 20 minutes. The solution was then kept at room temperature for 1.5 hours, after which 5 volumes of water were added. The dioxane was removed on the

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rotary evaporator, and the product was precipitated from the remaining aqueous s lution by addition of acetic acid. precipitate was filtered off and dissolved in a small volume of chloroform-methanol (1:1). Silica gel was added to the solution and the solvents removed in vacuo. The solids were applied to a silica gel column and the product eluted with methanol-ethyl acetate (1:4). This eluent did dissolve some silica gel which remained with the eluted product. (CD<sub>3</sub>OD, 10% dDMSO): δ 1.98ppm, 2.12ppm (2m, 4H, aliphatic methylenes),  $\delta$  2.72ppm, 3.06ppm (2m, 4H, anilinic methylenes), δ 3.56ppm (m, 4H, benzylic methylenes), δ 4.08ppm (s, 2H, bromoacetyl),  $\delta$  6.79ppm (dd, 1H, J<sub>1</sub>=9.2 Hz, J<sub>2</sub>=2.1 Hz, xanthene),  $\delta$  6.83ppm (s, 1H, xanthene),  $\delta$  6.90ppm (d, 1H, J=2.1 Hz, xanthene),  $\delta$  7.19ppm (d, 1H, J=9.2 Hz, xanthene),  $\delta$ 7.24ppm (d, 1H, J=8.4 Hz, phthalic), & 8.02ppm (dd, 1H,  $J_1=8.4$  Hz,  $J_2=1$  Hz, phthalic),  $\delta$  8.30ppm (d, 1H, J=1 Hz, phthalic).

For preparation of  $7\beta$ -(bromoacetamido)-3-[[[(5-rhodol-X-amido)methyl]thio]methyl]-3-cephem-4-carboxylic acid (20), 4.5mg (10μmol) 5-Rhodol-Xbromoacetamide (18) were dissolved in 0.5ml 250mM phosphate buffer adjusted to pH 7.7 and 0.5ml dimethylformamide. solution was decaygenated and 10mg (40 $\mu$ mol) 7 $\beta$ -amino-3-(thiomethyl)-3-cephem-4-carboxylic acid (8) prepared according to the literature procedure in  $100\mu l$  phosphate buffer was added in an argon atmosphere. The solution was kept for 2 hours at 30°C. Then the solvents were removed in vacuo and the residue dissolved in 1 ml water, from which the product was precipitated by addition of acetic acid. The precipitate was collected and the product purified by C10 reverse-phase chromatography with 0.1% trifluoroacetic acid in 35% methanol/water as eluent.

The above product (19) was dissolved in 1ml dioxane-water (1:1) with 20mg sodium bicarbonate.  $10\mu l$  Bromoacetyl bromide were added to the solution on ice. The solution was kept for another 1.5 hours at room temperature. 20mg sodium bicarbonate and  $10\mu l$  bromoacetyl bromide were added to the solution with ice cooling. After another 1.5 hours at room

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temperature the dioxane was removed on the rotary evaporator and the products were precipitated from the aqueous solution with 1M phosphoric acid and collected by c ntrifugation. solids were suspended in dilute aqueous bicarbonat solution and the undissolved particles removed by centrifugation and discarded. The product was precipitated with 1M phosphoric acid and purified by flash chromatography on silica gel with chloroform-methanol-acetic acid-water (55:15:4:2). procedure dissolved small amounts of silica gel.

Coupling of diacetyl 5-fluoresceinthiol (21) with  $7\beta$ -(bromoacetamido)-3-[[[(5-rhodol-Xamido)methyl]thio]methyl]-3-cephem-4-carboxylic acid (20) was effected as follows.  $7\beta$ -(Bromoacetamido)-3-[[[(5-rhodol-Xamido)methyl]thio]methyl]-3-cephem-4-carboxylic acid was reacted with a 50% excess of 5-fluoresceinthiol under argon with dimethylformamide-(250mM aqueous phosphate buffer pH 7.7) (1:1) as the solvent. The product was purified from excess fluoresceinthiol by repeated dissolution in methanol and precipitation in ethyl acetate.

Fig. 3 shows the fluorescence emission spectra of this FRET-cephalosporin in 50mM phosphate buffer pH 7 before and after treatment with  $\beta$ -lactamase. The low initial fluorescence is due to the stacking of the fluorophores, forming a ground state complex that is nonfluorescent. When one adds methanol to the solution this stacking can be disrupted and efficient fluorescence resonance energy transfer occurs.

## Example 4 (compound 25)

N-[resorufin-4-carbonyl]-N'-iodoacetyl-piperazine (Boehringer Mannheim) was attached to the cephalosporin as a FRET-acceptor for fluorescein. It is referred to as FCRE.

The FRET-cephalosporin FCRE (25) carrying fluorescein as the donor and resorufin as the quencher was made by the same procedure as the one carrying the rhodol-Xacceptor. The N-[resorufin-4-carbonyl]-N'-iodoacetylpiperazine (Boehringer Mannheim) was coupled to the free 3'-thiol of the cephalosporin followed by bromoacetylation and

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addition of the 5-fluoresceinthiol. In departure from the protocol, three equivalents of 5-flu rescein thi l were added, as the first quivalent instantaneously reduced the resorufin and formed unreactive diflu rescein-disulfide. Exposure to air reoxidized resorufin to the original dye.

 $\beta$ -Lactamase catalyzed hydrolysis of this compound generates two fluorescent fragments. Resorufin excitation and emission spectra are longer wavelength and narrower than the rhodol spectra, possibly affording better spectral separation between the uncleaved dye versus the products of enzymatic cleavage. But, as in the case of rhodol as the acceptor, in aqueous phosphate buffer the dyes stack and form a dark complex.  $\beta$ -Lactamase treatment disrupts the stacking and increases donor fluorescence (Fig. 4).

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## Example 5 (compound 7b)

- For synthesis of 2,4 dihydroxy-5-chlorobenzaldehyde, 21.7 g (0.15 Mol) 4-chlororesorcinol were dissolved in 150 ml dry diethyl ether and 27 g finely powdered zinc (II) cyanide and 0.5 g potassium chloride were added with stirring. The suspension was cooled on ice. A strong stream of hydrogen chloride gas was blown into the solution with vigorous stirring. After approximately 30 minutes the reactants were dissolved. The addition of hydrogen chloride gas was continued until it stopped being absorbed in the ether solution (approx. 1 hour). During this time a precipitate formed. The suspension was stirred for one additional hour on Then the solid was let to settle. The ethereal solution was poured from the solid. The solid was treated with 100 g of ice and heated to 100°C in a water bath. Upon cooling the product crystallized in shiny plates from the solution. They were removed by filtration on dried over potassium hydroxide.

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The yield was 15.9 g (0.092 Mol, 61%). 'H NMR (CDCl<sub>3</sub>): δ 6.23ppm (s, 1H, phenol), δ 6.62ppm (s, 1H, phenyl), δ 7.52ppm (s, 1H, phenyl), δ 9.69ppm (s, 1H, formyl), δ 11.25ppm (s, 1H, phenol).

To prepare 3-carboxy 6-chloro 7-hydroxy coumarin, 5.76 g (0.033 Mol) 2,4-dihydroxy-5-chlorobenzaldehyde and 7.2 g (0.069 Mol) malonic acid were dissolved in 5 ml warm pyridine. 75  $\mu$ l Aniline were stirred into the solution and the reaction let to stand at room temperature for 3 days. yellow solid that formed was broken into smaller pieces and 50 ml ethanol was added. The creamy suspension was filtered through a glass frit and the solid was washed three times with 1 N hydrochloric acid and then with water. Then the solid was stirred with 100 ml ethyl acetate, 150 ml ethanol and 10 ml half concentrated hydrochloric acid. The solvent volume was reduced in vacuo and the precipitate recovered by filtration, washed with diethyl ether and dried over phosphorous pentoxide. 4.97 g (0.021 Mol, 63%) of product was obtained as a white powder.  $^{1}$ H NMR (dDMSO):  $\delta$  6.95ppm (s, 1H),  $\delta$  8.02ppm  $(s, 1H), \delta 8.67ppm (s, 1H).$ 

To prepare 7-butyryloxy-3-carboxy-6-chlorocoumarin, 3.1 g (12.9 mMol) 3-carboxy-6-chloro-7-hydroxycoumarin were dissolved in 100 ml dioxane and treated with 5 ml butyric anhydride, 8 ml pyridine and 20 mg dimethyl aminopyridine at room temperature for two hours. The reaction solution was added with stirring to 300 ml heptane upon which a white precipitate formed. It was recovered by filtration and dissolved in 150 ml ethyl acetate. Undissolved material was removed by filtration and the filtrate extracted twice with 50 ml 1 N hydrochloric acid/brine (1:1) and then brine. solution was dried over anhydrous sodium sulfate. Evaporation in vacuo yielded 2.63 g (8.47 mMol, 66%) of product. H NMR (CDCl<sub>3</sub>):  $\delta$  1.08ppm (t, 3H, J = 7.4 Hz, butyric methyl),  $\delta$ 1.85ppm (m, 2H,  $J_1 = J_2 = 7.4$  Hz, butyric methylene),  $\delta$  2.68ppm (t, 2H, J = 7.4 Hz, butyric methylene),  $\delta$  7.37ppm (s, 1H, coumarin),  $\delta$  7.84ppm (s, 1H, coumarin),  $\delta$  8.86ppm (s, 1H, coumarin).

Preparation of 7-butyryloxy-3benzyloxycarbonylmethylaminocarbonyl-6-chlorocoumarin is effected as follows. 2.5 g (8.06 mMol) 7-Butyryloxy-3carboxy-6-chlorocoumarin, 2.36 g hydr xybenztriazole hydrate (16 mMol) and 1.67 g (8.1 mMol) dicyclohexyl carbodiimide were 5 dissolved in 30 ml dioxane. A toluene solution of 0benzylglycine [prepared by extraction of 3.4 g (10 mMol) benzylglycine tosyl salt with ethyl acetate - toluene saturated aqueous bicarbonate - water (1 : 1 : 1 : 1, 250 ml), drying of the organic phase with anhydrous sodium sulfate and .0 reduction of the solvent volume to 5 ml) was added dropwise to the coumarin solution. The reaction was kept at room temperature for 20 hours after which the precipitate was removed by filtration and washed extensively with ethylacetate and acetone. The combined solvent fractions were reduced to .5 50 ml on the rotatory evaporator upon which one volume of toluene was added and the volume further reduced to 30 ml. The precipitating product was recovered by filtration and dissolved in 200 ml chloroform - absolute ethanol (1 : 1). The solution was reduced to 50 ml on the rotatory evaporator :0 and the product filtered off and dried in vacuo yielding 1.29 g of the title product. Further reduction of the solvent volume yielded a second crop (0.64 g). Total yield: 1.93g (4.22 mMol, 52%). <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.08ppm (t, 3H, J = 7.4 Hz, butyric methyl),  $\delta$  1.84ppm (m, 2H,  $J_1 = J_2 = 7.4$  Hz, butyric 25 methylene),  $\delta$  2.66ppm (t, 2H, J = 7.4 Hz, butyric methylene),  $\delta$  4.29ppm (d, 2H, J = 5.5 Hz, glycine methylene),  $\delta$  5.24ppm (s, 2H, benzyl),  $\delta$  7.36ppm (s, 1H, coumarin),  $\delta$  7.38ppm (s, 5H, phenyl),  $\delta$  7.77ppm (s, 1H, coumarin),  $\delta$  8.83ppm (s, 1H, coumarin),  $\delta$  9.15ppm (t, 1H, J = 5.5 Hz, amide). 30 7-Butyryloxy-3-carboxymethylaminocarbonyl-6chlorocoumarin was prepared as follows. 920 mg (2 mMol) 7butyryloxy-3-benzyloxycarbonylmethylaminocarbonyl-6chlorocoumarin were dissolved in 50 ml dioxane. 100 mg palladium on carbon (10%) and 100  $\mu$ l acetic acid were added to 35 the solution and the suspension stirred vigorously in a hydrogen atmosphere at ambient pressure. After the uptake of

hydrogen seized the suspension was filtered. The product

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containing carbon was extracted five times with 25 ml boiling dioxane. The combined dioxane solutions w re let to cool upon which the product precipitated as a white powder. Reduction of the solvent to 20 ml precipitates more product. The remaining dioxane solution is heated to boiling and heptane is added until the solution becomes cloudy. The weights of the dried powders were 245 mg, 389 mg and 58 mg, totaling 692 mg (1.88 mMol, 94%) of white product. HNMR (dDMSO):  $\delta$  1.02ppm (t, 3H, J = 7.4 Hz, butyric methyl),  $\delta$  1.73ppm (m, 2H,  $J_1$ =  $J_2$  = 7.3 Hz, butyric methylene),  $\delta$  2.70ppm (t, 2H, J = 7.2 Hz, butyric methylene),  $\delta$  4.07ppm (d, 2H, J = 5.6 Hz, glycine methylene),  $\delta$  7.67ppm (s, 1H, coumarin),  $\delta$  8.35ppm (s, 1H, coumarin),  $\delta$  8.90ppm (s, 1H, coumarin),  $\delta$  9.00ppm (t, 1H, J = 5.6 Hz, amide).

Coupling of 7-Butyryloxy-3-15 carboxymethylaminocarbonyl-6-chlorocoumarin with 7-amino-3'chlorocephalosporanic acid benzhydryl ester was effected as follows. 368 mg (1 mMol) 7-Butyryloxy-3carboxymethylaminocarbonyl-6-chlorocoumarin, 270mg hydroxybenztriazole hydrate and 415 mg (1 mMol) 7-amino-3'-20 chloro cephalosporanic acid benzhydryl ester were suspended in 40 ml dioxane - acetonitrile (1 : 1). 260 mg (1.25 mMol) -dicyclohexylcarbodiimide in 5 ml acetonitrile were added and the suspension was stirred vigorously for 36 hours. precipitate was removed by filtration and the volume of the 25 solution reduced to 20 ml on the rotatory evaporator. Toluene was added and the volume reduced to 30 ml. With stirring 50 ml heptane was added and the suspension chilled on The precipitate was recovered by filtration. redissolved in 10 ml chloroform and the remaining undissolved 30 solids were filtered off. Addition of 2 volumes of heptane precipitated the title product which was collected and dried in vacuo and yielded 468 mg (0.64 mMol, 64%) off-white powder. <sup>1</sup>H NMR (CDCl<sub>3</sub>):  $\delta$  1.08ppm (t, 3H, J = 7.4 Hz, butyric methyl),  $\delta$  1.84ppm (m, 2H,  $J_1$ =  $J_2$  = 7.4 Hz, butyric methylene),  $\delta$ 35 2.66ppm (t, 2H, J = 7.4 Hz, butyric methylene),  $\delta$  3.54ppm (2d, 2H, J = 18.3 Hz, cephalosporin C-2),  $\delta$  4.24ppm (2d, 2H, J =5.8 Hz, cephalosporin 3 methylene),  $\delta$  4.37ppm (d, 2H, J = 3.8

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Hz, glycine methylene),  $\delta$  5.02ppm (d, 1H, J = 4.9 Hz, cephalosporin C-6),  $\delta$  5.89ppm (dd, 1H, J<sub>1</sub> = 9.0 Hz, J<sub>2</sub> = 5.0 Hz, cephalosporin C-7),  $\delta$  6.96ppm (s, 1H, benzhydryl),  $\delta$  7.30-7.45ppm (m, 12H, phenyl, coumarin, amide),  $\delta$  7.79ppm (s, 1H, coumarin),  $\delta$  8.84ppm (s, 1H, coumarin),  $\delta$  9.28ppm (t, 1H, J = 3.7 Hz, amide).

Coupling of the above product with 5fluoresceinthiol was effected as follows. 90 mg (0.2 mMol) 5mercaptofluorescein diacetate disulfide dimer were dissolved
in 10ml chloroform and treated with 25 µl tributyl phosphine
and 25 µl water in an argon atmosphere. The solution was kept
for 2 hours at ambient temperature and was then added to a
solution of 20 mg sodium bicarbonate, 25mg sodium iodide and
110 mg (0.15 mMol) of the above compound in 10 ml
dimethylformamide. After 4 hours the solvents were removed in
vacuo and the residue triturated with diethylether. The solid
was dissolved in ethyl acetate - acetonitrile (1:1). After
removal of the solvents the residue was triturated once more
with diethylether yielding 157 mg (0.13mMol, 88%) of a cream
colored powder product.

A sample of the above compound was treated with a large access of trifluoroacetic acid - anisole (1:1) at room temperature for 20 minutes. The reagents are removed in vacuo and the residue triturated with ether. High performance liquid chromatography of the solid in 45% aqueous acetonitrile containing 0.5% acetic acid gives a product in which the butyrate and the diphenylmethyl esters have been cleaved. It was purified by high performance liquid chromatography on a reverse phase C<sub>18</sub>-column using 45% aqueous acetonitrile containing 5% acetic acid as the eluent.

Deprotection of the fluorescein acetates in compound 27 was accomplished with sodium bicarbonate in methanol (room temperature, 30minutes) to provide the fluorescent enzyme substrate CCF2. It was purified by high performance liquid chromatography on a reverse phase C<sub>10</sub> - column using 35% aqueous acetonitrile containing 0.5% acetic acid as the eluent.

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Stirring of compound 27 with excess acetoxymethyl bromide in dry lutidine produced the membrane permeable derivative of the substrate (CCF2/ac<sub>2</sub>AM<sub>2</sub>). It was purified by high performance liquid chromatography on a reverse phase C<sub>18</sub> - column using 65% aqueous acetonitrile containing 0.5% acetic acid as the eluent. CCF2/ac<sub>2</sub>AM<sub>2</sub> is readily converted to CCF2 in the cells' cytoplasm.

Unlike in Examples 1-4, the donor and acceptor dyes in substrate CCF2 do not stack. The substrate is fully fluorescent in phosphate buffer and there is no formation of the "dark complex" (i.e., addition of methanol does not change the fluorescence spectrum of CCF2, except for the effect of dilution). This is due to the much smaller and more polar nature of the 7-hydroxycoumarin compared to that of the xanthene dyes (eosin, rhodamine, rhodol and resorufin) in Examples 1-4.

Fig. 5 illustrates the emission spectrum of compound CCF2 in 50 mmolar phosphate buffer pH 7.0 before and after  $\beta$ -lactamase cleavage of the  $\beta$ -lactam ring. In the intact substrate, efficient energy transfer occurs from the 7-

hydroxycoumarin moi ty t the fluorescein moiety. Excitati n of the substrate at 405 nm results in fluorescence emission at 515 nm (green) from the acceptor dye fluorescein. The energy transfer is disrupted when  $\beta$ -lactamase cleaves the  $\beta$ -lactam ring, thereby severing the link between the two dyes. Excitation of the products at 405 nm now results entirely in donor fluorescence emission at 448 nm (blue). The fluorescence emission from the donor moiety increases 25 fold upon  $\beta$ -lactam cleavage. The fluorescence at 515 nm is reduced by 3.5 fold, all of the remaining fluorescence originating from the 7-hydroxycoumarin as its emission spectrum extends into the green. Twenty-five-fold quenching of the donor in the substrate is equivalent to an efficiency of fluorescence energy transfer of 96%. This large fluorescence change upon  $\beta$ -lactam cleavage can readily be used to detect  $\beta$ -lactamase in the cytoplasm of living mammalian cells, as is reported in Examples 6 and 7.

The 7-hydroxycoumarin moiety in the cephalosporin was determined to have a fluorescence quantum efficiency in the absence of the acceptor of 98-100%. This value was determined by comparing the integral of the corrected fluorescence emission spectrum of the dye with that of a solution of 9-aminoacridine hydrochloride in water matched for absorbance at the excitation wavelength. It follows that 7hydroxycoumarin is an ideal donor dye, as virtually every photon absorbed by the dye undergoes fluorescence energy transfer to the acceptor.

#### Example 6

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Cells of the T-cell lymphoma line Jurkat were suspended in an isotonic saline solution (Hank's balanced salt solution) containing approximately  $10^{12} \beta$ -lactamase enzyme molecules per milliliter (approximately 1.7 nM; Penicillinase 205 TEM R', from Sigma) and 1 mg/ml rhodamine conjugated to dextran (40 kd) as a marker of loading. The suspension was passed through a syringe needle (30 gauge) four times. This causes transient, survivable disruptions of the cells' plasma membrane and allows entry of labeled dextran and  $\beta$ -lactamase.

W 96/30540 PCT/US96/04059

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Cells that had been successfully permeabiliz d contained  $\beta$ -lactamas and were red fluorescent when illuminated at the rhodamine excitation wavelength on a fluorescent microscope. The cells were incubated with  $5\mu M$  fluorogenic  $\beta$ -lactamase substrate, CCF2/ac<sub>2</sub>AM<sub>2</sub>, at room temperature for 30 minutes. Illumination with violet light (405 nm) revealed blue fluorescent and green fluorescent cells. All cells that had taken up the marker rhodamine-dextran appeared fluorescent blue, while cells devoid the enzyme appeared fluorescent green.

### Example 7

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Cells from cell lines of various mammalian origin were transiently transfected with a plasmid containing the RTEM  $\beta$ -lactamase gene under the control of a mammalian promotor. The gene encodes cytosolic  $\beta$ -lactamase lacking any signal sequence and is listed as SEQ. ID. 1. 10 to 48 hours after transfection cells were exposed to 5 µmol CCF2/ac2AM2 for 1 to 6 hours. In all cases fluorescent\_blue cells were detected on examination with a fluorescence microscope. Not a single blue fluorescent cell was ever detected in nontransfected control cells. To quantitate the fluorescence measurements the cells were first viewed through coumarin (450 DF 65) and then fluorescein (515 EFLP) emission filters and pictures were recorded with a charge couple device camera. The average pixel intensities of CCF2 loaded transfected cells (blue) and controls (green) at coumarin and fluorescein wavelength in COS-7 (Table 2) and CHO (Table 3) cells are summarized; values for 4 representative cells for each population are given. Thus, the substrate CCF2 revealed gene expression in single living mammalian cells.

Table 2

COS-7 (origin: SV40 transformed african green monkey kidney cells)

5	Table of pixel intensities	coumarin emission filter	fluorescein emission filter
•	Blue cell #1	27	20
	#2	34	23
	#3	31	31
	#4	22	33
0	Green cel#1	4	43
U		4	42
	#3	5	20
	#4	3	24

Table 3

CHO (origin: Chinese hamster ovary cells)

	Table of pixel intensities		coumarin emission filter	fluorescein emission filter
				112
	Blue cell	#1	98	
0	•	#2	70	113
·		#3	76	92
		#4	56	67
	Green cell	-#1	9	
		#2	9	102
5		#3	. <b>7</b>	101
•		#4	9	83

PCT/US96/04059

W 96/30540

Example 8

For preparation of 7-acetyloxy-3-(N-carboxymethyl-Nmethylaminocarbonyl)coumarin, 400 mg (1.6 mMol) 3-carboxy-7acetylcoumarin were refluxed with 4 ml thionyl chloride for 20 minutes. Excess thionyl chloride was removed by distillation and the residue (7-acetyloxy-3-chlorocarbonylcoumarin) stored in vacuo over potassium hydroxide pellets overnight. separate vessel 142.5 mg (1.6 mMol) sarcosine was dissolved in 1.05 ml (5.4 mMol) N-methyl trimethylsilyl trifluoroacetamide (MSTFA) and kept at room temperature for 16 hours. acetonitrile and 187  $\mu$ l (1.7 mMol) N-methylmorpholine were added and the solution was poured onto the solid 7-acetyloxy-3-chlorocarbonylcoumarin on ice. After stirring for 20 minutes on ice the solution was let to warm to room temperature. After 4 hours the solvents were removed in The residue was dissolved in methanol to deprotect the acid after which the solvent was removed in vacuo. The solid was dissolved in 30 ml ethylacetate-acetonitrile (2:1) and the solution extracted twice with an equal volume of 1 N hydrochloric acid and the with brine. The organic phase was

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dried ver anhydrous sodium sulfate. The solvent was removed in vacuo and the solid crystallized from boiling ethylacetat with addition of hexane. The yield was 316mg (1.0 mMol, 63%) of a white crystalline solid.

Coupling of 7-acetyloxy-3-(N-carboxymethyl-Nmethylaminocarbonyl) coumarin with 7-amino-3'chlorocephalosporanic acid benzhydryl ester was effected as follows. 62 mg (0.2 mMol) 7-acetyloxy-3-(N-carboxymethyl-Nmethylaminocarbonyl) coumarin was stirred with 1 ml dry methylene chloride to which 27 mg (0.2 mMol) hydroxybenztriazole and 41 mg dicyclohexyl carbodiimide had been added. A solution of 82.6 mg (0.2 mMol) 7-amino 3'chloro cephalosporanic acid benzhydryl ester in 1 ml methylene chloride was added dropwise over a period of 5 minutes. The reaction was stirred for 20 hours at room temperature after which the precipitate was removed by filtration. was evaporated in vacuo and the product extracted into methylene chloride. The solvent was removed once more and the residue dissolved in 1 ml ethyl acetate. Addition of three volumes of hexane precipitated the product which was recovered by centrifugation. The yield was 49.9 mg (70  $\mu$ Mol, 35%) of the product as a white powder.

Substituent in the above product to the 3'-iodo substituent was carried out as follows. 49.9 mg (70  $\mu$ Mol) of the above product was stirred with 52.5 mg sodium iodide (5 equivalents) in 1.2 ml dry methyl ethyl ketone at room temperature for 2 hours. The solvent was removed in vacuo and the residue dissolved in 2 ml ethyl acetate - methylene chloride (1:1) and extracted with cold 2% aqueous sodium thiosulfate solution, followed by two extractions with brine. The organic layer was dried over anhydrous sodium sulfate. The slightly orange powder (32 mg, 40  $\mu$ Mol, 57%) was used without further purification in the next reaction.

Coupling of above product with 5-mercaptofluorescein diacetate (product CCFlac, diphenylmethyl ester) was effected by dissolving 32 mg (40  $\mu$ Mol) of the iodo derivative in 0.4 ml dimethylformamide and 3.4 mg sodium bicarbonate added. 22 mg

WO 96/30540

69

(50  $\mu$ M 1) 5-mercaptofluorescein diac tate were dissolved in 0.3 ml deoxygenated dimethyl formamide and added to the iodo compound in an argon atmosphere. After 2 hours the solvent was removed in vacuo. The residue was suspended in methylene chloride - ethyl acetate (1:1). The organic solution was washed with water and dried over anhydrous sodium sulfate. The solvent was removed and the residue was triturated with ethyl ether hexane (1:1). Flash chromatography on 60 mesh silica gel with ethyl acetate - toluene (2:1) yielded 4.2 mg (4  $\mu$ Mol, 10%) of colorless product.

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Cleavage of the diphenylmethyl ester to give CCFlac, was effected as follows. 4 mg (4 µMol) of CCFlac, diphenylmethyl ester were treated with 200 µl trifluoroacetic acid - anisole - methylene chloride (10:1:10) on ice for 15 minutes. The reagents were removed in vacuo and the residue was dissolved in 0.5 ml ethyl acetate and the solvent evaporated in vacuo. The solid was triturated with ether and then dissolved in 0.5 ml methanol. Addition of the methanolic solution to 2 ml water precipitated the product. The product was recovered by centrifugation and dried in vacuo. The yield was 2mg (2 µMol, 50%) white solid. The compound was further purified by high performance liquid chromatography on a reverse phase C10-column using 55% aqueous acetonitrile containing 0.5% acetic acid as the eluent.

The fluorescence emission spectrum of CCF1 before and after  $\beta$ -lactamase cleavage (Fig. 6) was obtained from a sample of CCF1ac, that had been converted to CCF1 by treatment with orange peel acetyl esterase in 50 mmolar aqueous phosphate buffer pH 7.

Substrate CCF1 has similar fluorescence properties to substrate CCF2 in Example 5. In the intact substrate, efficient energy transfer occurs from the 7-hydroxycoumarin moiety to the fluorescein moiety. Excitation of the substrate at 390 nm results in fluorescence emission at 515 nm (green) from the acceptor dye fluorescein. The energy transfer is disrupted when  $\beta$ -lactamase cleaves the  $\beta$ -lactam ring, thereby severing the link between the two dyes. Excitation of the products at 390 nm now results entirely in donor fluorescence

PCT/US96/04059 WO 96/30540

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emissi n at 460 nm (blue). The fluorescence emission from the donor moiety increas s 25-fold upon  $\beta$ -lactam cleavage. fluorescence at 515 nm is reduced by 3-fold, all of the remaining fluorescence originating from the 7-hydroxycoumarin as its emission spectrum extends into the green. Twenty-fivefold quenching of the donor in the substrate is equivalent to an efficiency of fluorescence energy transfer of 96%. This large fluorescence change upon  $\beta$ -lactam cleavage can readily be used to detect  $\beta$ -lactamase in the cytoplasm of living mammalian cells, as is reported in Example 9.

#### Example 9

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Cells of the T-cell lymphoma line Jurkat were suspended in an isotonic saline solution (Hank's balanced salt solution) containing approximately  $10^{12}~\beta$ -lactamase enzyme molecules per milliliter (approximately 1.7 nM; Penicillinase 205 TEM R\*, from Sigma) and 1 mg/ml rhodamine conjugated to dextran (40 kd) as a marker of loading. The suspension was passed through a syringe needle (30 gauge) four times. causes transient, survivable disruptions of the cells' plasma membrane and allows entry of labeled dextran and  $\beta$ -lactamase. Cells which had been successfully permeabilized contained  $\beta$ lactamase and were red fluorescent when illuminated at the rhodamine excitation wavelength on a fluorescent microscope. The cells were incubated with 30 $\mu$ M fluorogenic  $\beta$ -lactamase substrate CCFlac, at room temperature for 30 minutes. Illumination with ultraviolet light (360 nm) revealed blue fluorescent and green fluorescent cells. All cells that had taken up the marker rhodamine-dextran appeared fluorescent blue, while cells devoid the enzyme appeared fluorescent green.

# Example 10:

The preferred membrane-permeable ester of CCF2 was prepared as follows:

Coupling of 5-fluoresceinthiol diacetate (5) and 7-amino-3'-chlorocephalosporanic acid benzhydryl ester was effected as follows. 450 mg (1mmol) 5-mercaptofluorescein diacetate disulfide dimer were dissolved in 30ml chloroform and treated with 50  $\mu$ l water and 125  $\mu$ l tributylphosphine in a nitrogen atmosphere which generated the free 5-fluoresceinthiol. 450 mg (1mmol) 7-Amino-3'-

W 96/30540 PCT/US96/04059

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chlorocephalosp ranic acid benzhydryl ester hydrochloride salt were dissolved in 10 ml acetonitrile with the h lp f 220 µl (2 mmol) N-methyl morpholine and the two solutions combin d after 30 minutes. One hour later the solvent volume was reduced to 5 ml and 50 ml carbon tetrachloride were added. The solvent volume was reduced to 15 ml and hexane added with stirring. The initial orange precipitate consisting mainly of N-methyl morpholine hydrochloride was removed by filtration. Upon further addition of two volumes of hexane 630 mg (0.76 mmol, 76%) white product was precipitated and collected.

The above product was coupled with 7-butyryloxy-3carboxymethyl aminocarbonyl-6-chlorocoumarin. 325 mg (0.88 7-Butyryloxy-3-carboxymethyl aminocarbonyl-6chlorocoumarin was dissolved in 15 ml hot dry dioxane. With rapid cooling 110  $\mu$ l (1 mmol) N-methyl morpholine in 1 ml dioxane and 115  $\mu$ l (0.9 mmol) isobutyl chloroformate in 8 ml methylene chloride were added. The reaction was kept at 0°C for 30 minutes after which 661 mg (0.8 mmol) of the above fluorescein-cephalosporin adduct in 7 ml dry methylene chloride were added. The solution was let to warm to room temperature and after 3 hours the solvents were removed in vaccuo. The residue was dissolved in 30 ml methylene chloride and twice extracted with one volume 10% aqueous acetic acid and once with water. The organic phase was dried over anhydrous sodium sulfate. Addition of 150 ml dry ethanol, reduction of the solvent volume to 50 ml and cooling to -20°C resulted in precipitation of the product (crude, 850mg). Purification was achieved by chromatography over silica gel with 25% ethyl acetate in toluene as the eluent. mmol, 26%) of white powderous product was collected.

Cleavage of the cephalosporin benzhydryl ester was accomplished by treatment with trifluoroacetic acid. 145 mg (0.12 mmol) of the above product was treated with trifluoroacetic acid / methylene chloride / anisole (10 / 10 / 1) at 0°C for 20 minutes. The reagents were removed in vacuo and the residue triturated with diisopropyl ether. The solid was dissolved in 1 ml dimethyl sulfoxide and the product precipitated by addition to 25 ml water. It was further

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purified on reverse phase  $C_{10}$  resin with a step gradient of 40 to 60% agu ous acetonitrile containing 0.5% acetic acid as the eluent yielding 74 mg (73  $\mu$ mol, 60%) white powder.

Protection of cephalosporin acid as membrane permeable acetoxymethyl ester was achieved as follows. 15 mg (15  $\mu$ mol) of the above product was dissolved in 250  $\mu$ l methylene chloride. 25  $\mu$ l Bromomethyl acetate and 50  $\mu$ l lutidine were added to the solution. The reaction was kept at ambient temperature for 7 hours after which the reagents were removed in vacuo. The residue was purified by flash chromatography on silica gel with ethyl acetate as the eluent. 15 mg (14  $\mu$ mol, 92%) white product was obtained. This compound, named CCF2/btAMac<sub>2</sub>, was used for intracellular detection of  $\beta$ -lactamase activity.

.5 Example 11:

Measurement of activation of an intracellular receptor: Activation of the intracellular glucocorticoid receptor was—measured by—its ability to upregulate the transcriptional activity of the glucocorticoid responsive element in the mouse mammary tumor virus promotor. This response to steroids was detected as increased intracellular  $\beta$ -lactamase activity on the substrate CCF2 causing an appropriate change in fluorescent signal.

The gene for plasmid encoded RTEM  $\beta$ -lactamase of Escherichia coli without a signal sequence (Sequence 1 of Fig. 7) was put under transcriptional control of the mouse mammary tumor virus promotor and introduced into a mammalian expression vector. This vector also carried the chloramphenical resistance marker for amplification of the plasmid in bacteria and the neomycin resistance marker for mammalian selection. It was introduced into baby hamster kidney (BHK) cells in culture using the calcium phosphate precipitation technique. Cells were then subjected to selection for stable integration of the plasmid into the cells' genome using the antibiotic G418. One of twenty clones was selected for its marked increase in  $\beta$ -lactamase expression following exposure to the steroid analog dexamethasone.

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The following describes the measurement of the increas in  $\beta$ -lactamase gene expression in this clone after addition of the agonist dexamethasone. Cells of the stable BHK cell clone G941 expressing  $\beta$ -lactamase under control f the glucocorticoid-inducible promotor were kept in the presence or absence of the agonists in the incubator at 37 °C. Flasks with cells were removed from the incubator at different intervals after agonist addition and the cells transferred into Hank's balanced salt solution containing 10 µmolar CCF2/btAMac2 . This compound becomes converted to the  $\beta$ lactamase accessible fluorescent substrate CCF2 by endogenous cytoplasmic esterases. Ten minutes later the cell supernatant containing CCF2/btAMac, was removed. 30 Minutes later the cells were imaged with a cooled CCD camera mounted on an epifluorescence microscope. Fluorescence measurements were taken with violet excitation light (filter 400DF15) and with blue (filter 450DF65) and green (filter 535DF45) emission filters. A ratio of blue versus green emission intensities was determined. The ratio is a measure of how much substrate has been converted to product. Using a 40x objective, 4 fields with approximately 60 cells each were imaged at each time point. The results show a significant increase in the ratio of fluorescent intensities reflective of increasing  $\beta$ lactamase expression and production.

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	time in presence of 1 $\mu$ M dexamethasone	0.0 hours	1.0 hours	2.0 hours	hours
	average ratio of fluorescence intensities 450DF65 / 535DF45	0.21 +/-	0.38 +/- 0.05	0.42 +/- 0.07	0.47 +/- 0.08

#### Example 12:

Measurement of cell surface receptor activation and intracellular signaling via second-messenger responsive

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elements: Activation of cell surface receptors leads to a change in intracellular messenger concentrations which in turn modulates intracellular transcription factor activity. In lymphocytes, an increase the intracellular concentration of the messenger ion calcium leads to the activation of the nuclear factor of activated T-lymphocytes (NFAT). This event increases transcription at promoters containing the NFAT-recognition site. An increase in calcium levels alone is sufficient to markedly increase transcription of a reporter gene such as  $\beta$ -lactamase regulated when it is put under transcriptional control of a promotor containing a trimer of NFAT sites.

The murine T-lymphocyte cell line B3Z was transiently cotransfected with two plasmids. One plasmid contained the  $\beta$ -adrenergic receptor, which localizes at the cells' surface, under the transcriptional control of the strong and constitutively active cytomegalovirus (CMV) promoter. The other plasmid contained the bacterial RTEM etalactamase gene from Escherichia coli modified for improved mammalian expression (sequence ID # 3 , with optimum mammalian Kozak sequence,  $\beta$ -globin leader sequence, pre-sequence removed) under the transcriptional control of a promotor containing a trimer of NFAT sites. The plasmids were introduced into cells using electroporation. 5x106 cells in 0.5 ml electroporation buffer were electroporated in the presence of 10  $\mu g$  each of both plasmids using the Biorad Gene Pulser (250V, 960  $\mu$ F, 16  $\mu$ sec). Twenty-four hours after transfection, cells were either incubated in the presence or absence of the  $\beta$ -adrenergic agonist isoproterenol (10  $\mu$ molar) for 5 hours. The supernatant was removed and replaced with Hank's balanced salt solution containing 10  $\mu$ molar CCF2/btAMac2. After 20 minutes at room temperature cells were washed with fresh buffer and viewed with the fluorescence microscope. 4% of isoproterenol treated cells appeared fluorescent blue (excitation filter 400DF15, emission filter 435 nm longpass) while no blue fluorescent cells were detectable in the control population (absence of agonist). Maximal stimulation with 2  $\mu M$  ionomycin and 50 ng/ml phorbol

ester for 5 hours resulted in 20% blue fluorescent cells in the p pulation.

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# Example 13:

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 $\beta$ -Lactamases from different microorganisms were modified for use as reporter enzymes in eukaryotic cells, preferably mammalian. The bacterial gene for these enzymes includes a N-terminal pre-sequence (first 23 amino acids of Sequence 2 of Fig. 7.) that targets the enzyme to the extracellular space. Following translocation a pre-sequence peptidase cleaves the 23 amino acid pre-sequence releasing the mature  $\beta$ -lactamase enzyme. RTEM  $\beta$ -lactamase from Escherichia coli including its bacterial pre-sequence (Sequence 2 of Fig. 7) was put into a mammalian expression vector under the control of the mouse mammary tumor virus promotor. construct was introduced into baby hamster kidney cells using the standard calcium phosphate precipitation technique.  $\beta$ -lactamase activity was found in the cell culture medium; no activity could be detected in the cell pellet. The amount of eta-lactamase activity in the medium was steroid dependent. Cells that had been in the presence of 1  $\mu M$  dexamethasone for 36 hours prior to the measurement produced threefold more enzyme than control. This makes the  $\beta$ -lactamase with its bacterial pre-sequence (Sequences 2 of Fig. 7) useful for an extracellular assay of mammalian reporter gene activity.

A preferred use of the  $\beta$ -lactamase reporter is where the enzyme is produced and retained in the cell cytoplasm. Therefore the bacterial signal sequence was removed and replaced by ATG (methionine) as the new translational start site in three modified RTEM  $\beta$ -lactamase genes (Sequences 1, 3, and 4 of Fig. 7). In order to increase expression of the  $\beta$ lactamases in mammalian cells, the RTEM eta-lactamases of Sequence 3 and 4 of Fig. 7 were constructed with altered ribosome binding sites optimized for mammalian expression [Kozak, M., J. Cell Biol. 108: 229-241 (1989)]. For increased compatibility with the mammalian translation machinery,  $\beta$ lactamase of sequence ID #3 was inserted at the end of an untranslated mammalian eta-globin leader sequence.

these novel DNA sequences encoding nov 1  $\beta$ -lactamases wer inserted into mammalian expression vectors with the cytomegalovirus prom tor controlling their transcripti n. Mammalian cells in tissue culture (Hela, COS-7, CHO, BHK) wer transfected transiently with the plasmids using the standard lipofectin technique. Two to five days after transfection, the cells were incubated with the membrane-permeant derivative, CCF2/btAMac2, of the fluorescent substrate CCF2 to assay functional expression of the enzyme. 5-20% of cells transfected with plasmids containing cDNA Sequences 2, 3 and 4 of Fig. 7 showed a conversion of green to blue fluorescence indicating cleavage of the intracellularly trapped substrate by expressed  $\beta$ -lactamase. By contrast, in untransfected or mock transfected controls, all cells showed the green fluorescence of uncleaved CCF2; no blue-fluorescing cells were observed, confirming the absence of any endogenous eta-lactamase activity.

The gene for Bacillus licheniformis eta-lactamase was isolated from total Bacillus licheniformis DNA by use of the polymerase chain reaction. The oligonucleotide primers removed the  $\beta$ -lactamase secretion sequence and generated the DNA sequence ID # 5. This gene was inserted in a pCDNA3 mammalian expression vector under the transcriptional control of the constitutively active cytomegalovirus promoter. HeLa cells were transfected with 10  $\mu g$  of plasmid per 25 cm<sup>2</sup> culture dish using lipofectin. 5 days after transfection, cells were tested for functional expression of  $\beta$ -lactamase by incubating them in the presence of 100  $\mu$ molar CCF2/btAMac<sub>2</sub> and visual inspection with the epifluorescence microscope. 30-40% of cells showed blue fluorescence, whereas only greenfluorescing cells, no blue-fluorescing cells were detectable in untransfected controls. In transient transfections, it is typical for <50% of the cells to become transfected.

### 5 Example 14

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A plasmid was constructed with  $\beta$ -lactamase of sequence ID 3 (figure 7) under control of yeast elongation factor EF-lalpha enhancer and promoter. This plasmid was

coinjected tog ther with the potassium salt of substrate CCF2 (compound 7b) into z brafish embryos at the single cell stage. As control, mbryos were injected with the potassium salt of substrate CCF2 alone. After three hours, the embryos were viewed with an epifluorescence microscope using violet excitation light (filter 400DF15) and a 435 nm longpass emission filter. Embryos that had received plasmid DNA fluoresced blue while controls fluoresced green.

#### 10 Example 15

The  $\beta$ -lactamase gene of sequence ID 3 was cloned into a Drosophila transformation vector under the control of the glass promotor and injected into wild-type Drosophila embryos. As control, the  $\beta$ -lactamase gene was inserted in the wrong orientation. Drosophila embryos were germline-15 transformed using P element-mediated transformation. transformations and all subsequent fly manipulations were performed using standard techniques [Karess, R.E. and Rubin, ---G.M., Cell 38,--135,-(1984)]. Omatidia-of late stage transformed pupe were transsected and dissociated to single 20 cells. The cells were incubated in buffer with 40  $\mu molar$ CCF2/btAMac2 (compound 34) for 20 minutes, washed and viewed with an epifluorescence microscope (excitation filter 400DF15, emission filter 435 nm long pass). Omatidia cells from flyes transformed with the  $\beta$ -lactamase gene in the proper 25 orientation fluoresced blue, while omatidia cells containing the gene in the wrong orientation fluoresced green.

# Example 16

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In certain embodiments the compound of this invention can be any of the following compounds.

wherein

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 $R^{\gamma}$  is selected from the group consisting of H, Cl, and Br;

 $R^{\mathbf{x}}$  is selected from the group consisting of H, and methyl;

wherein R' and R'1 are independently selected from the group consisting of -C(0)alk, -CH2OC(0)alk, -CH2SC(0)alk, -CH2OC(O)Oalk, lower acyloxy-alpha-benzyl, and deltabutyrolactonyl; wherein alk is lower alkyl of 1 to 4 carbon atoms, and membrane-permeant fluorogenic derivatives thereof,

R' is 1-(acyloxy)alkyl.

Another example of the compound is:

wherein  $R^{z}$  and  $R^{z1}$  are independently selected from the group 10 consisting of -C(0)alk, -CH2OC(0)alk, -CH2SC(0)alk, -CH2OC(O)alk, lower acyloxy-alpha-benzyl, and deltabutyrolactonyl; wherein alk is lower alkyl of 1 to 4 carbon atoms.

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Another example of the compound is:

A final example of the compound is:

PCT/US96/04059

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All publicati ns and patent documents cited in this application are incorporated by referenc in their entirety for all purposes to the same extent as if each individual publication or patent document were so individually denoted.

The present invention provides novel substrates for beta-lactamase, beta-lactamases and methods for their use. While specific examples have been provided, the above description is illustrative and not restrictive. Many variations of the invention will become apparent to those skilled in the art upon review of this specification. The scope of the invention should, therefore, be determined not with reference to the above description, but instead should be determined with reference to the appended claims along with their full scope of equivalents.

All publications and patent documents cited in this application are incorporated by reference in their entirety for all purposes to the same extent as if each individual publication or patent document were so individually denoted.

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# WHAT IS CLAIMED IS:

A compound of the formula (I):

$$X$$
  $-z$   $N$   $A$   $Z^{1}$   $Y$   $CO_2R^{*}$ 

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wherein:
          one of X and Y is a fluorescent donor moiety or a
membrane-permeant derivative thereof, and the other is a
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quencher moiety, an acceptor fluorophore moiety or a membrane-

permeant derivative thereof;

R' is selected from the group consisting of H, lower alkyl,  $(CH_2)_nOH$ ,  $(CH_2)_nCOOR*$ , and =NOJ, in which n is 0 or an integer from 1 to 5 and J is H, Me, CH2COOH, CHMeCOOH, and

CMe2COOH; 0 1

R\* is selected from the group consisting of H, physiologically acceptable metal and ammonium cations, -

CHR<sup>2</sup>OCO(CH<sub>2</sub>)<sub>n</sub>CH<sub>3</sub>, -CHR<sup>2</sup>OCOC(CH<sub>3</sub>)<sub>3</sub>, acylthiomethyl, acyloxy-alpha-

benzyl, delta-butyrolactonyl, methoxycarbonyloxymethyl,

phenyl, methylsulphinylmethyl, betamorpholinoethyl,

dialkylaminoethyl, dialkylaminocarbonyloxymethyl, in which R2

is selected from the group consisting of H and lower alkyl;

A is selected from the group consisting of S, O, SO, SO<sub>2</sub> and CH<sub>2</sub>;

Z' is a linker for X; and

Z\* is a linker for Y.

- The compound of claim 1, wherein Z' is selected 1 from the group consisting of a direct bond, 2  $-(CH_2)_nCONR^2(CH_2)_n-, -(CH_2)_nNR^2CO(CH_2)_n-, -(CH_2)_nNR^3CONR^2(CH_2)_n-, -(CH_2)_nNR^2(CH_2)_n-, -(CH_2)_n-, -(CH_2)_n-, -(CH_2)_n-, -(CH_2)_n-, -(CH_2)_n-, -(CH_2$ 3  $(CH_2)_nNR^3CSNR^2(CH_2)_n$ ,  $-(CH_2)_nCONR^3(CH_2)_pCONR^2(CH_2)_n$ -,  $-(CH_2)_n$ -, -4
- $(CH_2)_nNR^3CO(CH_2)_pS(CH_2)_n-, -(CH_2)_nS(CH_2)_n-, -(CH_2)_nO(CH_2)_n-,$ 5

 $-(CH_2)_nNR^2(CH_2)_n-, -(CH_2)_nSO_2NR^2(CH_2)_n-, -(CH_2)_nCO_2(CH_2)_n-,$ 5

- wherein R2 and n are as previously defined; R3 is selected from 7 the group consisting of hydrogen and lower alkyl; and each of
- m and p is independently selected from the group consisting of 8 9
- integers from 0 to 4. .0
- The compound of claim 1 wherein Z" is selected 1 from the group consisting of a direct bond to a heteroatom 2 selected from the group consisting of O, N and S in the 3 chromophore, 4
- $-O(CH_2)_n^-$ ,  $-S(CH_2)_n^-$ ,  $-NR^2(CH_2)_n^-$ ,  $-N^*R^2_2(CH_2)_n^-$ , 5
- $-\mathsf{OCONR}^2\left(\mathsf{CH}_2\right)_{\mathfrak{n}^-}, \ \ -\mathsf{O}_2\mathsf{C}\left(\mathsf{CH}_2\right)_{\mathfrak{n}^-}, \ \ -\mathsf{SCSNR}^2\left(\mathsf{CH}_2\right)_{\mathfrak{n}^-}, \ \ \mathsf{SCSO}\left(\mathsf{CH}_2\right)_{\mathfrak{n}^-},$ 6
- $-S(CH_2)_nCONR^2(CH_2)_n$ ,  $-S(CH_2)_nNR^2CO(CH_2)_n$ , and

- wherein R2 and n are as previously defined; and m is an
- integer from 0 to 4. 9
- The compound of claim 1 wherein R' is selected 1 from the group consisting of H and methyl.
- The compound of claim 4 wherein R'is H. 5. 1
- The compound of claim 1 wherein R' is selected 1
- from the group consisting of H and acetoxymethyl. 2

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- The compound of claim 1 wherein R2 is H. 7.
- The compound of claim 1 wherein A is -S-. 8.
- The compound of claim 1 wherein one of X and Y 9. is a membrane-permeant derivative of the fluorescent donor moiety, or a membrane-permeant derivative of the quencher moiety or acceptor fluorophore moiety, and wherein at least one of X and Y contains at least one acylated aromatic hydroxyl, acylated amine, or alkylated aromatic hydroxyl wherein the acyl group contains 1 to 5 carbon atoms and wherein the alkyl group is selected from the group consisting -CH<sub>2</sub>OC(O)alk, -CH<sub>2</sub>SC(O)alk, -CH<sub>2</sub>OC(O)Oalk, lower acyloxy-alphabenzyl, and deltabutyrolactonyl; wherein alk is lower alkyl of 1 1 to 4 carbon atoms. 2
- The compound of claim 9 wherein at least one of 1 X and Y contains at least one acylated aromatic hydroxyl, 2 wherein the acyl group is either acetyl, n-propionyl, or n-3 butyryl. 4
- The compound of claim 1 wherein said donor is a 1 coumarin of formulae II-VII and said quencher or acceptor is 2 selected from the group consisting of fluoresceins, rhodols, 3 and rhodamines of formulae VIII-XII, XLVII, and XLVII.
- The compound of claim 11 wherein said donor is 1 selected from the group consisting of coumarins of formulae 2 II-VII and fluoresceins of formula VIII.
- The compound of claim 1 wherein said quencher 1 or acceptor is selected from the group consisting of 2 fluoresceins, rhodols, and rhodamines of formulae VIII-XII. 3
- The compound of claim 13 wherein said donor is 1 a fluorescein of formula VIII and said quencher or acceptor is 2

W 96/30540 PCT/US96/04<del>059</del>

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- select d from the group consisting of rhodols and rh damines of formula VIII-XII.
- 15. The compound of claim 12 wherein said coumarin
  2 is selected from the group consisting of 7-hydroxycoumarin and
  3 7-hydroxy-6-chlorocoumarin and said fluorescein is selected
  4 from the group consisting of fluorescein and
- dichlorofluorescein.
- 5 dichlorofluorescein.
- 16. The compound of claim 13 wherein said donor is 2 a fluorescein and said quencher or acceptor is selected from 3 the group consisting of an eosin or a tetrachlorofluorescein 4 of formula VIII in which R<sup>a</sup>, R<sup>b</sup>, R<sup>c</sup>, and R<sup>d</sup> are Br or Cl.
- 1 17. The compound of claim 13 wherein said quencher or acceptor is a rhodol of formulae VIII, IX, or XI.
- 18. The compound of claim 13 wherein said quencher or acceptor is a rhodamine of formulae VIII, X, or XII.
- 19. The compound of claim 9 that is membranepermeant derivatives wherein at least one of X and Y contains
  an acetoxymethyl group on an aromatic hydroxyl group.
- 1 20. The compound of claim 2 wherein Z' is 2 -(CH<sub>2</sub>)<sub>n</sub>CONR<sup>2</sup>(CH<sub>2</sub>)<sub>n</sub>-.
- 1 21. The compound of claim 20 wherein n and m are 0.
- 1 22. The compound of claim 20 wherein R<sup>2</sup> is H.
- 1 23. The compound of claim 3 wherein  $Z^*$  is  $-S(CH_2)_n$ -.
- 1 24. The compound of claim 23 wherein n is 0.

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25. The compound of claim 12 of the formula:

wherein

 $R^{Y}$  is selected from the group consisting of H, Cl, and Br;

 $R^{x}$  is selected from the group consisting of H, and methyl;

wherein R<sup>2</sup> and R<sup>21</sup> are independently selected from the group consisting of -C(O)alk, -CH<sub>2</sub>OC(O)alk, -CH<sub>2</sub>SC(O)alk, -CH<sub>2</sub>OC(O)alk, lower acyloxy-alpha-benzyl, and deltabutyrolactonyl; wherein alk is lower alkyl of 1 to 4 carbon atoms, and membrane-permeant fluorogenic derivatives thereof,

R' is 1-(acyloxy)alkyl

1 26. The compound of claim 25 wherein R<sup>2</sup> and R<sup>21</sup> are selected from the group consisting of acetyl, butyryl, and acetoxymethyl.

27. The compound of claim 12 of the formula:

2 wherein

 $R^{Y}$  is selected from the group consisting of H, Cl, and Br; and

- Rx is selected from the group consisting of H, and methyl.
- 1 28. The compound of claim 14 of the formula (11):

- 2 and membrane-permeant fluorogenic derivatives thereof.
- 1 29. The compound of claim 16 of the formula (16):

- wherein R<sup>2</sup> and R<sup>21</sup> are independently selected from the group
- consisting of -C(0)alk, -CH2OC(0)alk, -CH2SC(0)alk, -
- 4 CH<sub>2</sub>OC(0) alk, lower acyloxy-alpha-benzyl, and
- 5 deltabutyrolactonyl; wherein alk is lower alkyl of 1 to 4
- 6 carbon atoms.

30. The compound of claim 14 of the formula (22):

2 and membrane-permeant fluorogenic derivatives thereof.

2

31. The compound f claim 1 of the formula (25):

- 2 and membrane-permeant fluorogenic derivatives thereof.
- 1 32. The compound of claim 27 wherein  $R^{Y}$  is Cl and  $R^{X}$  2 is H. (CCF2 or 76).
  - 33. The compound of claim 15 that is

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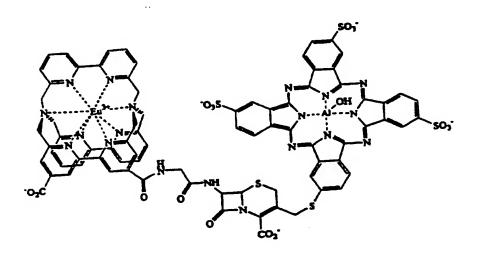
3

1

1 34. The compound f claim 26 wherein
2 R<sup>2</sup> and R<sup>2</sup> are acetoxymethyl;
3 R<sup>2</sup> is H; and
4 each R<sup>21</sup> is acetyl.

1 35. The compound of claim 27 wherein 2 R<sup>Y</sup> is H; and R<sup>X</sup> is methyl (CCF1 7a).

- 36. The compound of claim 1 wherein said donor is selected from the group consisting of fluorescent europium and terbium complexes.
  - 37. The compound of claim 36 wherein said donor is selected from the group consisting of europium and terbium tris-(bipyridine) cryptands and related dyes.
    - 38. The compound of claim 1 wherein said quencher or acceptor is selected from the group consisting of phthalocyanines and related dyes.
- 1 39. The compound of claim 37 wherein said quencher 2 or acceptor is selected from the group consisting of 3 phthalocyanines and related dyes.
  - 40. The compound of claim 39 that is



The compound of claim 10 wherein said coumarin 41. 1 is a compound of formula (III): 2

III

wherein 3

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E is selected from the group consisting of H, OH, ORk, and NR9Rh; 5

T is selected from the group consisting of O and NRk;

 $R^{\bullet}$  and  $R^{\flat}$  are independently selected from the group consisting of an attachment point, H, halogen, and lower alkyl;

 $R^{\alpha}$ ,  $R^{h}$ , and  $R^{k}$  are independently selected from the group consisting of an attachment point, H, lower alkyl, and -CH2 (CH2) na;

12 Ri is selected from the group consisting of an attachment 13 point, H, halogen, lower-alkyl, CN, CF, phenyl, CO,H, and L4 CONR PRb; and 15

a is selected from the group consisting of H or an L6 attachment point; and membrane-permeant fluorogenic 17 derivatives thereof. 18

42. The compound of claim 11 wherein said acceptor 1 or quencher is a compound of formula (VIII): 2

wherein 3

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E is selected from the group consisting of H, OH, ORk, and RgRh;

G is selected from the group consisting of O and N°RSRh;

Q' is selected from the group consisting of 0,  $CH_2$ , C(CH<sub>3</sub>)<sub>2</sub>, and NR<sup>9</sup>;

L

1

 $R^a$ ,  $R^b$ ,  $R^c$ , and  $R^d$  are independently sel cted from the group consisting f an attachment point, H, halog n, and low r alkyl;

R\* is selected from the group consisting of an attachment point, H, lower alkyl, (CH<sub>2</sub>)<sub>a</sub>CO<sub>2</sub>H, (CH<sub>2</sub>)<sub>a</sub>CHaCO<sub>2</sub>H, CHa(CH<sub>2</sub>)<sub>a</sub>CO<sub>2</sub>H, (CH<sub>2</sub>) COa, CH=CHCOa;

 $R^g$ ,  $R^h$ ,  $R^k$  are independently selected from the group consisting of an attachment point, H, lower alkyl, and - $CH_2(CH_2)_na;$ 

n is an integer of from 0 to 5;

- a and a' are independently selected from the group consisting of H or an attachment point; and membrane-permeant fluorogenic derivatives thereof.
- The compound of claim 42 wherein said donor is L selected from the group consisting of coumarins of formulae 2 II-VII, and membrane-permeant fluorogenic derivatives thereof. 3
  - The compound of claim 43 wherein
  - Q' is C(CMe2)2; and
- G is O; and membrane-permeant fluorogenic derivatives 3 thereof.

1 45. The compound f claim 43 selected from the group consisting of:

46. A method for determining whether a sample contains  $\beta$ -lactamase activity comprising:

contacting the sample with a compound of claim 1, which exhibits fluorescence resonance energy transfer when the compound is excited;

exciting the compound; and determining the degree of fluorescence resonance energy transfer in the sample, whereby a degree of fluorescence resonance energy transfer that is lower than an expected amount indicates the presence of  $\beta$ -lactamase activity.

amount of an enzyme in a sample wherein determining the degree of fluorescence resonance energy transfer in the sample comprises determining the degree at a first and second time after contacting the sample with the substrate, and determining the difference in the degree of fluorescence resonance energy transfer, whereby the difference in the degree of fluorescence in the degree of fluorescence resonance energy transfer reflects the amount of enzyme in the sample.

W 96/30540 PCT/US96/04059

94

48. A recombinant nucleic acid molecule comprising
 expression c ntr l sequences adapted for function in a
 vertebrate cell and operably linked to a nucleotide sequence
 coding for the expression of a β-lactamase.

- 1 49. The recombinant nucleic acid molecule of claim 2 48 wherein the nucleotide sequence encodes a class A  $\beta$ 3 lactamase.
- 50. The recombinant nucleic acid molecule of claim
  49 further comprising mammalian Kozak sequences for the
  ribosome binding site.
- 51. The recombinant nucleic acid molecule of claim
  50 wherein the nucleotide sequence is Sequence 3 or Sequence 4
  51. The recombinant nucleic acid molecule of claim
  52. The recombinant nucleic acid molecule of claim
  53. The recombinant nucleic acid molecule of claim
  54. The recombinant nucleic acid molecule of claim
  55. The recombinant nucleic acid molecule of claim
  56. The recombinant nucleic acid molecule of claim
  57. The recombinant nucleic acid molecule of claim
  58. The recombinant nucleic acid molecule of claim
  59. The recombinant nucleic acid molecule of claim
  50. The recombinant nucleic acid molecule of claim
  51. The recombinant nucleic acid molecule of claim
  51. The recombinant nucleic acid molecule of claim
  52. The recombinant nucleic acid molecule of claim
  53. The recombinant nucleic acid molecule of claim
  54. The recombinant nucleic acid molecule of claim
  55. The recombinant nucleic acid molecule of claim
  56. The recombinant nucleic acid molecule of claim
  57. The recombinant nucleic acid molecule of claim
  58. The recombinant nucleic acid molecule of claim
  58. The recombinant nucleic acid molecule of claim
  59. The recombinant nucleic acid molecule of
- 52. The recombinant nucleic acid molecule of claim
  49 wherein the nucleotide sequence is Sequence 1 or Sequence 5
  of Figure 7.
- 53. The recombinant nucleic acid molecule of claim
  48 wherein the expression control sequences are inducible
  expression control sequences.
- 1 54. The recombinant nucleic acid molecule of claim 2 48 wherein the expression control sequences are constitutively 3 active expression control sequences.
- 1 55. The recombinant nucleic acid molecule of claim
  2 48 wherein the expression control sequences include the c-fos
  3 promoter or the c-jun promoter.
- 1 56. The recombinant nucleic acid molecule of claim
  2 48 wherein the expression control sequences include cyclic
  3 AMP-responsive elements, phorbol ester response elements,
  4 serum response element, or Nuclear Factor of Activated T-cells
  5 response elements.

WO 96/30540

95

PCT/US96/04059

- The rec mbinant nucleic acid molecule of claim
  wherein the expression control sequences are responsive to
  substances that modulate cell-surface receptors.
- 1 58. The recombinant nucleic acid molecule of claim 2 48 wherein the expression control sequences are responsive to 3 substances that modulate intra-cellular receptors.
- 1 59. The recombinant nucleic acid molecule of claim 2 48 wherein the  $\beta$ -lactamase comprises a signal sequence for extracellular secretion.
- 1 60. The recombinant nucleic acid molecule of claim 2 59 wherein the nucleotide sequence is Sequence 2 of Figure 7.
- 1 61. The recombinant nucleic acid molecule of claim 2 48 wherein the  $\beta$ -lactamase is a cytosolic  $\beta$ -lactamase.
- 1 62. The recombinant nucleic acid molecule of claim
  2 61 wherein the nucleotide sequence is Sequence 1, Sequence 3,
  3 Sequence 4, or Sequence 5 of Fig. 7.
- 63. A recombinant nucleic acid molecule comprising
  expression control sequences adapted for function in a
  eukaryotic cell and operably linked to a nucleotide sequence
  coding for the expression of a cytosolic β-lactamase.
- 1 64. The recombinant nucleic acid molecule of claim 2 63 wherein the nucleotide sequence encodes a class A  $\beta$ 3 lactamase.
- 1 65. The recombinant nucleic acid molecule of claim
  2 64 wherein the nucleotide sequence is Sequence 1, Sequence 3,
  3 Sequence 4 or Sequence 5 of Figure 7.

W 96/30540 PCT/US96/04<del>059</del>

96

1 66. The recombinant nucleic acid molecule of claim
2 63 wherein the expression control sequences are inducible
3 expression control sequences.

- 1 66. The recombinant nucleic acid molecule of claim
  2 63 wherein the expression control sequences are constitutively
  3 active expression control sequences.
- 1 67. The recombinant nucleic acid molecule of claim
  2 63 wherein the expression control sequences include the c-fos
  3 promoter or the c-jun promoter.
- 1 68. The recombinant nucleic acid molecule of claim
  2 63 wherein the expression control sequences include cyclic
  3 AMP-responsive elements, phorbol ester response elements,
  4 serum response element, or Nuclear Factor of Activated T-cells
  5 response elements.
- 1 69. The recombinant nucleic acid molecule of claim
  2 63 wherein the expression control sequences are responsive to
  3 substances that modulate cell-surface receptors.
- 70. The recombinant nucleic acid molecule of claim
  2 63 wherein the expression control sequences are responsive to
  3 substances that modulate intra-cellular receptors.
- 71. A mammalian host cell transfected with a
  recombinant nucleic acid molecule further comprising
  expression control sequences adapted for function in a
  mammalian cell and operably linked to the nucleotide sequence
  coding for the expression of a β-lactamase.
- 72. A method of determining the amount of β-lactamase activity in a cell comprising the steps of:
  providing a host cell transfected with a
  recombinant nucleic acid molecule comprising expression
  control sequences operatively linked to nucleic acid sequences
  coding for the expression of a β-lactamase;

amount of  $\beta$ -lactamase activity.

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- contacting a sample comprising the cell or an 7 extract of the cell or conditioned medium with a substrate f r 8  $\beta$ -lactamase; and 9 determining the amount of substrate cleaved, 0 whereby the amount of substrate cleaved is related to the
- A method for monitoring the expression of a 73. 1 gene operably linked to a set of expression control sequences 2 comprising: 3

providing a eukaryotic host cell transfected with a recombinant nucleic acid molecule comprising expression control sequences operatively linked to nucleic acid sequences coding for the expression of a  $\beta$ -lactamase, except if the eukaryote is a fungus, wherein the  $\beta$ -lactamase is a cytosolic B-lactamase;

9 contacting a sample comprising the cell or an .0 extract of the cell or conditioned medium with a substrate for .1 β-lactamase; and

.2 determining the amount of substrate cleaved, ٤3 whereby the amount of substrate cleaved is related to the 14 amount of  $\beta$ -lactamase activity. 15

- The method of claim 73 wherein the eukaryotic 1 cell is mammalian. 2
- The method of claim 73 wherein the sample 1 comprises an extract from the cell, with or without the conditioned medium. 3
- The method of claim 73 wherein the sample 1 comprises the cell. 2
- The method of claim 73 wherein the substrate is 1 a compound of claim 1 and the step of determining the amount 2 of substrate cleaved comprises exciting the compound; and determining the degree of fluorescence resonance energy transfer in the sample, whereby a degree of fluorescence 5

PCT/US96/04059 W 96/30540

98

- resonance energy transfer that is lower than an expected 6 amount indicates the presence of  $\beta$ -lactamase activity. 7
- The method of claim 73 wherein the nucleotide 1 sequence encodes a class A  $\beta$ -lactamase. 2
- The method of claim 78 wherein the nucleotide 1 sequence is Sequence 1, Sequence 2, Sequence 3, Sequence 4 or 2 Sequence 5 of Figure 7. 3
- The method of claim 73 wherein the expression 80. 1 control sequences are inducible expression control sequences. 2
- The method of claim 73 wherein the cell is an 1 animal cell and the  $\beta$ -lactamase is a cytosolic  $\beta$ -lactamase. 2
- A method for determining whether a test 1 compound alters the expression of a gene operably linked to a 2 set of expression control sequences comprising: 3 providing a cell transfected with a recombinant 4 nucleic acid construct comprising the expression control 5 6

sequences operably linked to nucleic acid sequences coding for the expression of a -\beta-lactamase except if the eukaryote is a 7 fungus, wherein the  $\beta$ -lactamase is a cytosolic  $\beta$ -lactamase; 8

contacting the cell with the test compound; 9 contacting a sample comprising the cell or an 10

extract of the cell with a  $\beta$ -lactamase substrate; and

11 determining the amount of substrate cleaved, 12

whereby the amount of substrate cleaved is related to the 13 amount of  $\beta$ -lactamase activity. 14

- 83. The method of claim 82 wherein the eukaryotic 1 host cell is mammalian. 2
- 84. The method of claim 82 wherein the sample 1 comprises an extract from the cell. 2
- 85. The method of claim 82 comprises the cell. 1

W 96/30540 PCT/US96/04059

99

1 86. The method of claim 82 wherein the substrate is 2 a compound of claim 1 and the step f determining the amount 3 of substrate cleaved comprises exciting the compound; and 4 determining the degree of fluorescence resonance energy 5 transfer in the sample, whereby a degree of fluorescence 6 resonance energy transfer that is lower than an expected 7 amount indicates the presence of β-lactamase activity.

- 1 87. The method of claim 82 wherein the nucleotide 2 sequence encodes a class A  $\beta$ -lactamase.
- 1 88. The method of claim 87 wherein the nucleotide 2 sequence is Sequence 1, Sequence 2, Sequence 3, Sequence 4 or 3 Sequence 5 of Figure 7.
- 1 89. The method of claim 85 wherein the expression 2 control sequences are inducible expression control sequences.
- 1 90. The method of claim 85 wherein the cell is an 2 animal cell and the  $\beta$ -lactamase is a cytosolic  $\beta$ -lactamase.

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- 91. The method of claim 82 wherein the expression of  $\beta$ -lactamase is operably regulated by activation, or inhibition of activation, of an intracellular receptor, a cell surface receptor, or a component of an intracellular signaling pathway.
- 92. A method of clonal selection comprising:
   providing cells transfected with a recombinant
  nucleic acid molecule comprising the expression control
  sequences operably linked to nucleic acid sequences coding for
  the expression of a cytosolic β-lactamase;
   contacting the cells with a substance that activates

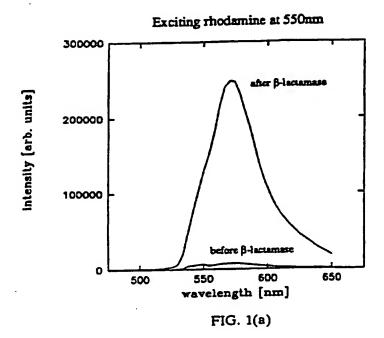
contacting the cells with a substance that activates or inhibits the activation of the expression control sequences;

8 sequences;
 9 contacting the cells with a compound of claim 9
 10 which is converted into a substrate;

WO 96/30540 PCT/US96/04<del>059</del>

100

.1	determining whether substrate is cleaved within ach
.2	individual c 11, wher by cleavage refl cts $\beta$ -lactamase
.3	activity:
.4	selecting and propagating those cells with a
.5	selected level of $eta$ -lactamase activity.
.6	93. The method of claim 92 further comprising the
.7	steps of:
.8	culturing selected cells in the absence of
.9	activator for a time sufficient for cleaved substrate to be
.o	substantially lost from the cells and for $\beta$ -lactamse levels to
:0 :1	return to unactivated levels;
	incubating the selected cells with a compound
?2	of claim 9 which is converted into a substrate; and
23	selecting cells that have not substantially
24	•
25	cleaved the substrate.



RCF fluorescence emission

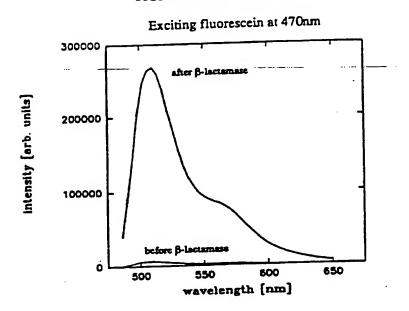
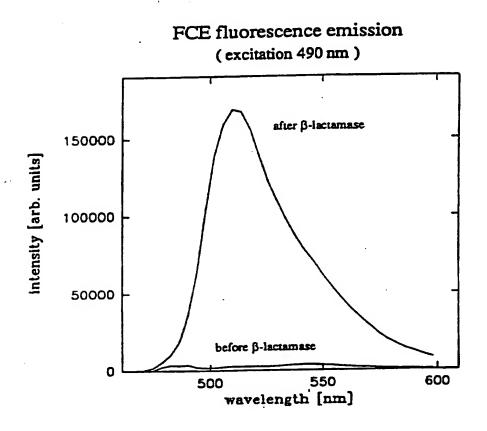


FIG. 1(b)

FIG. 2



WO 96/30540 PCT/US96/04059

3/17

FIG. 3

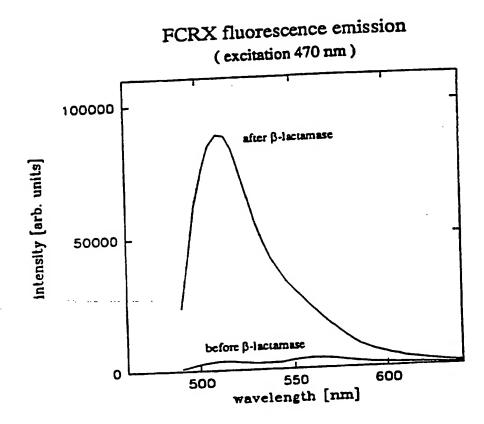
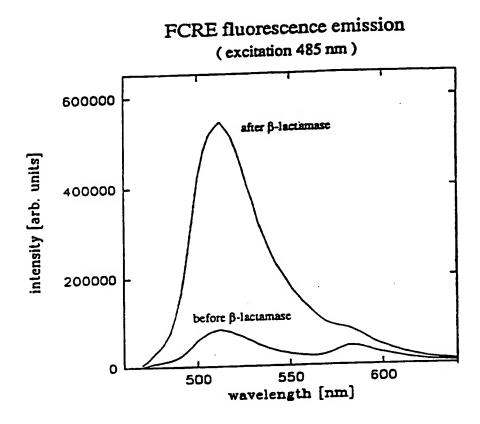


FIG. 4



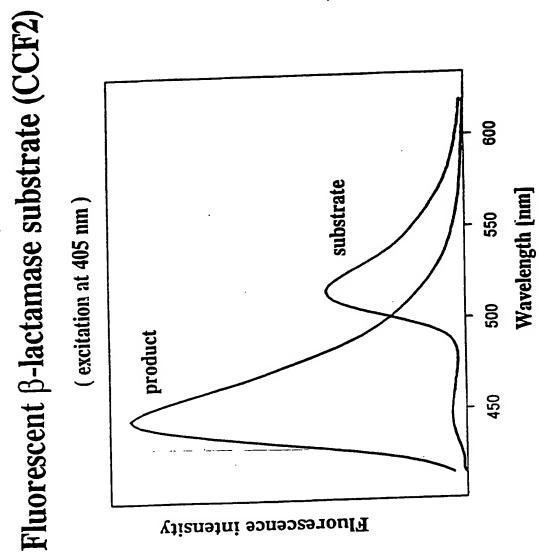


FIG. 5

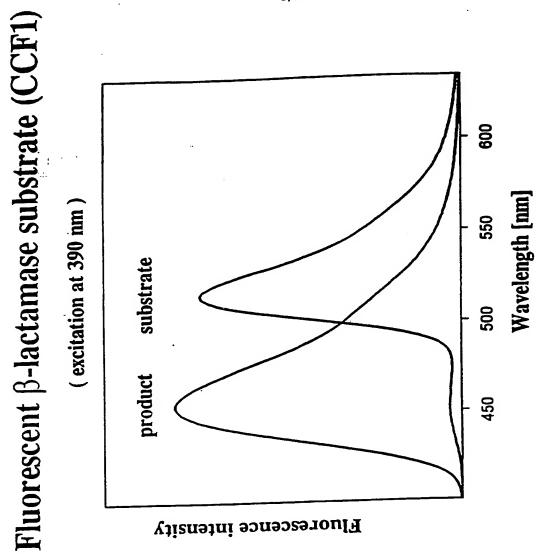


FIG. 6

FIGURE 7A

Se q ID	parent $\beta$ - lactamase gene and reference	modification	mammalian expression vector	location of expressio n
#	Escherichia coli RTEM Kadonaga et al.	signal sequence replaced by: ATG TGA	pMAM-neo glucocortic oid- inducible	cytoplasm ic
# 2	Escherichia coli RTEM Kadonaga et al.	wild type secreted enzyme 2 changes in pre- sequence: ser 2 arg , ala 23 gly	pMAM-neo glucocortic oid- inducible	secreted extracell ularly
3	Escherichia coli RTEM	β-globin up stream leader: AAGCTTTTTGCAGAAGCTCA GAATAAACGCAACTTTCCG Kozak sequence: GGTACCACCATGG signal sequence replaced by: ATG GGG	pCDNA 3 CMV promotor and pZEO SV40 promotor	cytoplasm ic
# 4	Escherichia coli RTEM	Kozak sequence: GGTACCACCATGG signal sequence replaced by: ATG GAC (GAC replaces CAT)	pMAM-neo glucocortic oid inducible	cytoplasm ic
# 5	Bacillus licheniform is 749/C Neugebauer et al.	signal sequence removed, new N-terminal ATG	pCDNA 3 CMV promotor	cytoplasm ic

J. Kadonaga et al., J. Biol. Chem., 259: 2149 (1984)
K. Neugebauer et al., Nucleic Acid Res., 9: 2577 (1981)

# FIGURE 7B Sequence no. 1: range 1 to 795 E. coli RTEM as modified by Kadonaga et al (1984)

```
ATG AGT CAC CCA GAA ACG CTG GTG AAA GTA AAA GAT GCT GAA GAT CAG TTG
Met Ser His Pro Glu Thr Leu Val Lys Val Lys Asp Ala Glu Asp Gln Leu
                                                                       100
                                                         90
                                           80
                            70
GGT GCA CGA GTG GGT TAC ATC GAA CTG GAT CTC AAC AGC GGT AAG ATC CTT
Gly Ala Arg Val Gly Tyr Ile Glu Leu Asp Leu Asn Ser Gly Lys Ile Leu
110 120 130 140 150
GAG AGT TIT CGC CCC GAA GAA CGT TIT CCA ATG ATG AGC ACT TIT AAA GTT
Glu Ser Phe Arg Pro Glu Glu Arg Phe Pro Met Met Ser Thr Phe Lys Val
                                                     190
                                     180
                      170
       160
CTG CTA TGT GGC GCG GTA TTA TCC CGT GTT GAC GCC GGG CAA GAG CAA CTC Leu Leu Cys Gly Ala Val Leu Ser Arg Val Asp Ala Gly Gln Glu Gln Leu
                                                   240
                    220
     210
GGT CGC CGC ATA CAC TAT TCT CAG AAT GAC TTG GTT GAG TAC TCA CCA GTC Gly Arg Arg Ile His Tyr Ser Gln Asn Asp Leu Val Glu Tyr Ser Pro Val
                                                 290
                                  280
                  270
ACA GAA AAG CAT CTT ACG GAT GGC ATG ACA GTA AGA GAA TTA TGC AGT GCT
Thr Glu Lys His Leu Thr Asp Gly Met Thr Val Arg Glu Leu Cys Ser Ala
                                                 340
                                330
               320
GCC ATA ACC ATG AGT GAT AAC ACT GCG GCC AAC TTA CTT CTG ACA ACG ATC
Ala Ile Thr Met Ser Asp Asn Thr Ala Ala Asn Leu Leu Leu Thr Thr Ile
                                                               400
                               380
                                               390
                370
360
GGA GGA CCG AAG GAG CTA ACC GCT TTT TTG CAC AAC ATG GGG GAT CAT GTA
Gly Gly Pro Lys Glu Leu Thr Ala Phe Leu His Asn Met Gly Asp His Val
                                                              450
                                               440
               420
                               430
                •
ACT CGC CTT GAT CGT TGG GAA CCG GAG CTG AAT GAA GCC ATA CCA AAC GAC
Thr Arg Leu Asp Arg Trp Glu Pro Glu Leu Asn Glu Ala Ile Pro Asn Asp
                                            490
                            480
               470
GAG CGT GAC ACC ACG ATG CCT GCA GCA ATG GCA ACA ACG TTG CGC AAA CTA
Glu Arg Asp Thr Thr Met Pro Ala Ala Met Ala Thr Thr Leu Arg Lys Leu
```

# FIGURE 7C

			E21			54	10			550			560	)	
520			530			-	•	•		•		•	•		•
TTA ACT		•				-	-	TCC	CGG	CAA	CAA	TTA	ATA	GAC	TGG
TTA ACT Leu Thr	GGC	GAA	CTA	CII	ACT	CIA	CCI	100	200	C1-	Cln	1.411	Tle	Ago	Tro
Leu Thr	Gly	Glu	Leu	Leu	Thr	Lea	770	Ser	MIG.	9111	GIII		637		
570			580	)		59	90		•	500	_		010	,	•
	•		*		*	*		•		•					-
ATG GAG	-	CAT	AAA	CTT	GCA	GGA	CCA	CII	CIG	CGC	TCG	GCC	CII	CCG	GCT
ATG GAG Met Glu	GCG	GAI	T.	17-1	Ala	Gly	Pro	Leu	Leu	Arg	Ser	Ala	Leu	Pro	YIS
	Ala	ABP	TAB	vai	YIG	640			650	) _		66	60		
620			30					•	•		•		•	•	
GGC TGG	•		*	•				-	COT	CNG	CCT	CCC	TCT	CGC	GGT
GGC TGG	TTT	ATT	GCT	GAT	AAA	TCT	GGA	GCC	661	GAG	2	Clu	Sat	Ara	Glv
GGC TGG Gly Trp	Phe	Ile	Ala	Asp	Lys	Ser	Gly	YIS	327	GIU	Arg	GIY	, Ser	AL 9	0-,
670	• • • • •	68	0	_	69	90			700			/10	,	_	
	•		-	*		•			•		*	*			
ATC ATT		CCX	CTG	CCC	CCA	GAT	GGT	AAG	CCC	TCC	CGT	ATC	GTA	GIT	ATC
ATC ATT	GLA	GCA	C10	21	D=0	Agn	Glv	Lvs	Pro	Ser	Arg	Ile	Val	Val	Ile
Ile Ile	Ala	ALA	Leu	GIA	710	40		-,-	750		_	760			
720		731	-	_			_		•			•		*	•
TAC ACG		*		•	-			~~~	C3.3	CCA	3 3 T	NGA	CAG	ATC	GCT
TAC ACG	ACG	GGG	AGT	CAG	GCA	ACT	ATG	GAI	GAA	Z	VVI	YOU	Gln	TIA	Ala
TAC ACG	ACG Thr	GGG	AGT Ser	CAG Gln	GCA Ala	ACT Thr	ATG Met	Asp	Glu	Arg	Asn	Arg	Gln	Ile	Ala
Tyr Thr	ACG Thr	_Gly _GGG	AGT Ser 780	CAG Gln	GCA Ala	THE	ATG Met 90	Asp	Glu	Arg	Asn	Arg	Gln	Ile	Ala
Tyr Thr	Thr	_Gly	Ser 780 ★	GID	*	7	90 •	*		Arg	Asn	Arg	Gln	Ile	Ala
Tyr Thr	Thr	_Gly	Ser 780 ★	GID	*	7	90 •	*		Arg	Asn	Arg	Gln	Ile	Ala
Tyr Thr	Thr	GCC • Gly	Ser 780 + TCA	CTG	BIA +	7. AAG	90 CAT	TGG	<b>02</b>	Arg	Asn	Arg	Gln	Ile	Ala

PCT/US96/04059

#### 10/17

#### FIGURE 7D

Sequence no. 2: range 1 to 858 Wild-type secreted RTEM enzyme with Ser2-Arg, Ala23-Gly

```
50
                                       30
ATG AGA ATT CAA CAT TTC CGT GTC GCC CTT ATT CCC TTT TTT GCG GCA TTT
Met Arg Ile Gln His Phe Arg Val Ala Leu Ile Pro Phe Phe Ala Ala Phe
                                                   90
                        70
                                      80
         60
TGC CTT CCT GTT TTT GGT CAC CCA GAA ACG CTG GTG AAA GTA AAA GAT GCT
Cys Leu Pro Val Phe Gly His Pro Glu Thr Leu Val Lys Val Lys Asp Ala
                      120
                                    130
        110
GAA GAT CAG TTG GGT GCA CGA GTG GGT TAC ATC GAA CTG GAT CTC AAC AGC
Glu Asp Gln Leu Gly Ala Arg Val Gly Tyr Ile Glu Leu Asp Leu Asn Ser
                                  180
                                                 190
                     170
GGT AAG ATC CTT GAG AGT TTT CGC CCC GAA GAA CGT TTT CCA ATG ATG AGC Gly Lys Ile Leu Glu Ser Phe Arg Pro Glu Glu Arg Phe Pro Met Met Ser
                                               240
                                 230
                   220
     210
ACT TTT AAA GTT CTG CTA TGT GGC GCG GTA TTA TCC CGT GTT GAC GCC GGG
Thr Phe Lys Val Leu Leu Cys Gly Ala Val Leu Ser Arg Val Asp Ala Gly
                                              290
                                280
                  270
    260
CAA GAG CAA CTC GGT CGC CGC ATA CAC TAT TCT CAG AAT GAC TTG GTT GAG
Gln Glu Gln Leu Gly Arg Arg Ile His Tyr Ser Gln Asn Asp Leu Val Glu
                                                           350
                                             340
                              330
                 320
   310
TAC TCA CCA GTC ACA GAA AAG CAT CTT ACG GAT GGC ATG ACA GTA AGA GAA
Tyr Ser Pro Val Thr Glu Lys His Leu Thr Asp Gly Met Thr Val Arg Glu
                                           390
                             380
                370
TTA TGC AGT GCT GCC ATA ACC ATG AGT GAT AAC ACT GCG GCC AAC TTA CTT
Leu Cys Ser Ala Ala Ile Thr Met Ser Asp Asn Thr Ala Ala Asn Leu Leu
                                                        450
                                          440
                             430
              420
CTG ACA ACG ATC GGA GGA CCG AAG GAG CTA ACC GCT TTT TTG CAC AAC ATG
Leu Thr Thr Ile Gly Gly Pro Lys Glu Leu Thr Ala Phe Leu His Asn Met
                                                                     510
             470
                           480
460
GGG GAT CAT GTA ACT CGC CTT GAT CGT TGG GAA CCG GAG CTG AAT GAA GCC
Gly Asp His Val Thr Arg Leu Asp Arg Trp Glu Pro Glu Leu Asn Glu Ala
```

# FIGURE 7E

						530			540	)		5	550			560
			520	_				•	•		•		•	•		•
	*		CNC	GAG	ССТ	GAC	ACC	ACG	ATG	CCT	GCA	GCA	ATG	GCA	ACA	ACG
ATA	CCA	AAC	GAC	97	\	200	The	Thr	Met	Pro	Ala	Ala	Met	Ala	Thr	Thr
Ile	Pro	Asn	ABP	GIU	AIG.	580	1111	****	590			600	)		•	510
		570	,		•	-	•		•		*			•		*
	*	*								N CT	СТА	CCT	TCC	CGG	CAA	CAA
TTG	CGC	AAA	CTA	TTA	ACT	GGC	GAA	CIA	CII	WCI	CIA	33-		220	Gln	CAA
Leu	Arg	LVS	Leu	Leu	Thr	Gly	Glu	Ten	Den	Thr	Leu	~~~	SEI	ALY	660	
	3	620			630	0			540			650			661	,
•				*	•		•		*	*		*				
	242	CAC	TCC	ATG	GAG	GCG	GAT	AAA	GTT	GCA	GGA	CCA	CIT	CIG	CGC	TCG
TTA	WIW	GAC	700	Wat	Glu	Ala	Agn	LVS	Val	Ala	Gly	Pro	Leu	Leu	Arg	Ser
Leu			TTP	MEC	680	774	····	690	n			700			710	
	•	570			980		_	•		•		*	•		•	
*	•	•	•				-	-	CAT		T)	CCA	GCC	CCT	GAG	CGT
GCC	CII	CCG	GCT	GGC	TGG	T-I-I	ATT	GCI	GAI	7	50-	Clar	27.	Gly	Glu	CGT
Ala	Leu	Pro	Ala	Gly	Trp	Phe	Ile	Ala	ABP	Lys	361		AT G	G1,	750	Arg
	72			-	730			740			750	J			, 60	•
•			•		•	•		•		•						
-	TOTAL	CGC	CCT	ATC	ATT	GCA	GCA	CTG	GGG	CCA	GAT	GGT	AAG	CCC	TCC	CGT
000	201	200	C):-	Tle	TIA	Ala	Ala	Leu	Gly	Pro	Asp	Gly	Lys	Pro	Ser	Arg
GIA		Arg	GIY	78	~	7124		790			800	-		81	0	
	770				_			•	•		•		•	*		•
	•		•					3 CT	CAG	CCA	على ت	ATG	GAT	GAA	CGA	AAT
ATC	GTA	GTT	ATC	TAC	ACG	ACG	-3-0	AGI	21-	212	The	Mot	Aen	Glu	Ara	AAT Asn
Ile	Val	Val	Ile	Tyr	Thr	Thr	GLY	Ser	GIII	YIG	1111	ME	n.p		3	Asn
	820			830			84	0		,	B50	_				
	•	•		•		•	•		*							
AGA	CAG	ATC	GCT	GAG	ATA	GGT	GCC	TCA	CTG	ATT	AAG	CAT	166			
1.00		-1	33-	03	T 3 A	Glv	212	Ser	Leu	Ile	Lys	His	Trp			

## FIGURE 7F

Sequence no. 3: range 1 to 795 RTEM enzyme with  $\beta$ -globin upstream leader, mammalian Kozak equence, replacement of signal sequence by Met Gly

```
\GCTTTTTGCAGAAGCTCAGAATAAACGCAACTTTCCGGGTACCACC
                                                                     50
                                       30
                          20
           10
IG GGG CAC CCA GAA ACG CTG GTG AAA GTA AAA GAT GCT GAA GAT CAG TTG GGT GCA
et Gly His Pro Glu Thr Leu Val Lys Val Lys Asp Ala Glu Asp Gln Leu Gly Ala
                                            90
                             80
                70
IA GTG GGT TAC ATC GAA CTG GAT CTC AAC AGC GGT AAG ATC CTT GAG AGT
g Val Gly Tyr Ile Glu Leu Asp Leu Asn Ser Gly Lys Ile Leu Glu Ser
                                          140
            120
                          130
TT CGC CCC GAA GAA CGT TTT CCA ATG ATG AGC ACT TTT AAA GTT CTG CTA Le Arg Pro Glu Glu Arg Phe Pro Met Met Ser Thr Phe Lys Val Leu Leu
                                                         200
                          180
                                          190
            170
FT GGC GCG GTA TTA TCC CGT GAT GAC GCC GGG CAA GAG CAA CTC GGT CGC
             .
's Gly Ala Val Leu Ser Arg Ile Asp Ala Gly Gln Glu Gln Leu Gly Arg
                                                  250
                                      240
                         230
C ATA CAC TAT TOT CAG AAT GAC TTG GTT GAG TAC TCA CCA GTC ACA GAA IG Ile His Tyr Ser Gln Asn Asp Leu Val Glu Tyr Ser Pro Val Thr Glu
                                                    300
                                     290
                       280
LG CAT CIT ACG GAT GGC ATG ACA GTA AGA GAA TTA TGC AGT GCT GCC ATA
/8 His Leu Thr Asp Gly Met Thr Val Arg Glu Leu Cys Ser Ala Ala Ile
                                                   350
                                     340
                     330
       320
IC ATG AGT GAT AAC ACT GCG GCC AAC TTA CTT CTG ACA ACG ATC GGA GGA
ir Met Ser Asp Asn Thr Ala Ala Asn Leu Leu Leu Thr Thr Ile Gly Gly
                                                 400
                                  390
                    380
      370
EG AAG GAG CTA ACC GCT TTT TTG CAC AAC ATG GGG GAT CAT GTA ACT CGC
O Lys Glu Leu Thr Ala Phe Leu His Asn Met Gly Asp His Val Thr Arg
                                                450
                                                                460
                                 440
                   430
    420
T GAT CAT TGG GAA CCG GAG CTG AAT GAA GCC ATA CCA AAC GAC GAG CGT
au Asp His Trp Glu Pro Glu Leu Asn Glu Ala Ile Pro Asn Asp Glu Arg
```

# FIGURE 7G

			•	480	1		4	190			500			510	)	
	470					•		*	*		•		•	•		•
GAC	•				~	-	3.770	CCA	ACA	ACG	TTG	CGC	AAA	CTA	TTA	ACT
GAC	ACC	ACG	ATG	CCI	GIA	GCA.	MIG	272	The	Thr	Len	Arg	LVB	Leu	Leu	Thr
Asp	Thr	Thr	Met	Pro	Val	ATA	MCC	WIG	1111		550	5	_,_	560		
	20			530			54(	3		•	,,,,	•				•
	*	•		*		*	•				-		030	TOC	377	GAG
GGC	GAA	CTA	CIT	ACT	CTA	GCT	TCC	CGG	CAA	CAA	TIA	ATA	GAC	100	Mar	Clu
Gly	Glu	Leu	Leu	Thr	Leu	Ala	Ser	Arg	Gln	Gln	Leu	Ile	Asp	TIP	HEL	GIU
				80			590	_		601	)			510		
570		_	_		•		•		•	*		•		*	•	
			C-TP-TT	CCA	CCA	CCA	CTT	CTG	CGC	TCG	GCC	CII	CCG	GCT	GGC	TGG
GCG	GAT	AAA	GII	33-	Clu	DTO	Leu	Leu	Ara	Ser	Ala	Leu	Pro	Ala	Gly	Trp
	qaA	Lys	Val	MIG	GTA	FIU	640		3	650			660	0		
620			630	)		•	940			*		•	•		•	
•		•	•						C3.C	CCT	ccc	TCT	CGC	GGT	ATC	ATT
TIT	ATT	GCT	GAT	AAA	TCT	GGA	GCC	COT	GAG	201	63.0	Car	Ara	Glv	Tle	ATT
Phe	Ile	Ala	Asp	Lys	Ser	Gly	Ala	GTÅ	GIU	w a	GIY	261	71	, ,		11e 720
670			680	)		6:	90			700					•	•
	•		•		•	•								N.E.C	TAC	ACC.
CCA	CCA	CTG	GGG	CCA	GAT	GGT	AAG	CCC	TCC	CGT	ATC	GIA	GIT	AIC	INC	ACG
27.	Ala	Leu	Glv	Pro	Asp	Gly	Lys	Pro	SET	~~	120				171	Thr . 770
V10	~~~		730		•	740			·-·7	50 ·			760		-	. //0
	•					•		•	•		•		*			
		3.07	CNG	GCA	ACT	ATG	GAT	GAA	CGA	AAT	AGA	CAG	ATC	GCT	GAG	ATA Ile
ACG	630	AGI	CAG	Ala	Thr	Met	Asp	Glu	Arg	Asn	Arg	Gln	Ile	Ala	Glu	Ile
Thr	GIA	ser	OT!	ALA		790					_					
		78	U	_		, 50	•									
	•			. —			TCC									
GGT	GCC	TCA	CTG	ATT	AAG	CAT	100									
Gly	Ala	Ser	Leu	Ile	Lys	H18	TP									

#### FIGURE 7H

Sequence no. 4: range 1 to 792 RTEM  $\beta$ -lactamase with mammalian Kozak sequence and replacement signal sequence by Met Asp

```
50
                                     30
G GAC CCA GAA ACG CTG GTG AAA GTA AAA GAT GCT GAA GAT CAG TTG GGT
t Asp Pro Glu Thr Leu Val Lys Val Lys Asp Ala Glu Asp Gln Leu Gly
                                                 90
                      70
                                    80
A CGA GTG GGT TAC ATC GAA CTG GAT CTC AAC AGC GGT AAG ATC CTT GAG
a Arg Val Gly Tyr Ile Glu Leu Asp Leu Asn Ser Gly Lys Ile Leu Glu
                                                             150
                                               140
                                  130
                   120
     110
T TTT CGC CCC GAA GAA CGT TTT CCA ATG ATG AGC ACT TTT AAA GTT CTG
r Phe Arg Pro Glu Glu Arg Phe Pro Met Met Ser Thr Phe Lys Val Leu
                                              190
                               180
                  170
A TGT GGC GCG GTA TTA TCC CGT ATT GAC GCC GGG CAA GAG CAA CTC GGT u Cys Gly Ala Val Leu Ser Arg Ile Asp Ala Gly Gln Glu Gln Leu Gly
                              230
                                            240
                220
  210
C CGC ATA CAC TAT TCT CAG AAT GAC TTG GTT GAG TAC TCA CCA GTC ACA
g Arg Ile His Tyr Ser Gln Asn Asp Leu Val Glu Tyr Ser Pro Val Thr
                                                         300
                                           290
                              280
               270
 260
A AAG CAT CTT ACG GAT GGC ATG ACA GTA AGA GAA TTA TGC AGT GCT GCC
u Lys His Leu Thr Asp Gly Met Thr Val Arg Glu Leu Cys Ser Ala Ala
                                          340
                           330
              320
310
'A ACC ATG AGT GAT AAC ACT GCG GCC AAC TTA CTT CTG ACA ACG ATC GGA
e Thr Met Ser Asp Asn Thr Ala Ala Asn Leu Leu Leu Thr Thr Ile Gly
                                        390
                          380
             370
IA CCG AAG GAG CTA ACC GCT TTT TTG CAC AAC ATG GGG GAT CAT GTA ACT
y Pro Lys Glu Leu Thr Ala Phe Leu His Asn Met Gly Asp His Val Thr
                                                      450
                                        440
                           430
            420
IC CTT GAT CAT TGG GAA CCG GAG CTG AAT GAA GCC ATA CCA AAC GAC GAG
g Leu Asp Arg Trp Glu Pro Glu Leu Asn Glu Ala Ile Pro Asn Asp Glu
                                                                   510
                                                     500
                                       490
                         480
           470
            ٠
IT GAC ACC ACG ATG CCT GTA GCA ATG GCA ACA ACG TTG CGC AAA CTA TTA
g Asp Thr Thr Met Pro Val Ala Met Ala Thr Thr Leu Arg Lys Leu Leu
```

PCT/US96/04059

# 15/17

## FIGURE 71

						530			540	)		5	550			560
			20					•	•		*		•	*		•
	•		*		2	AT"	GCT	TCC	CGG	CAA	CAA	TTA	ATA	GAC Asp	TGG	ATG
ACT	GGC	GAA	CIA	CII	701	7	210	Car	ATG	Gln	Gln	Leu	Ile	Asp	Trp	Met
Thr	Gly	Glu	Leu	Leu	Thr	Leu	VIG	562	590			600	3	Asp	•	510
	_	570	)			80			230		•			•		•
	*	•				*	*						CTT.	CCC	CCT	GGC
CAG	CCG	CAT	AAA	GTT	GCA	GGA	CCA	CII	CIG	CGC	TCG	GCC	CII	CCG Pro	31-	Gly
GAG	33.	3.00	T 1/0	Val	Ala	Glv	Pro	Leu	Leu	Arg	Ser	Ala	Leu	Pro	WIG	GTY
GIU	ATA	WBD	Dyb	464	630	`,			540			650			660	,
		620					•		•	•		•		*	•	
*		•		•			-	CCC	CCT	CAG	CT	GGG	TCT	CGC	GGT	ATC
TGG	TTT	ATT	GCT	GAT	AAA	TCI	GGA	33-	23		y ~~	Glv	Ser	Arg	Glv	Ile
TID	Phe	Ile	Ala	Asp	Lys	Ser	GIA	VT.	. 023	GIU	wid	3027	561	• 5	710	
		570		_	680			69	0			700			7.20	
							•			•		•	-			
-				~~~	CCA	CAT	CCT	AAG	CCC	TCC	CGT	ATC	GTA	GTT Val	ATC	TAC
ATT	GCA	GCA	CIG	53	200	700	Gly	LVS	Pro	Ser	Arq	Ile	Val	Val	Ile	TYI
Ile	Ala	Ala	Leu	GIA	PIO	WRD	GIY	240			75	n		Val	760	
	72				730			740		_	, ,	•	•		•	•
•	•		•		*	•						202	CNC	ATC	CCT	GAG
ACC.	ACG	GGG	AGT	CAG	GCA	ACT	ATG	GAT	GAA	CGA	AAT	AGA	CAG	ATC Ile	23-	Glu
WF-	The	Gly	CAT	Gln	Ala	Thr	Met	Asp	Glu	Arg	Asn	Arg	GIN	Ile	WIG	614
ini	1111	GIY	367	78	^			790								
	770			70	•	•		•								
	*	_				330	ChT	TCC								
ATA	GGT	GCC	TCA	CTG	ATT	AAG	TIÁ.	7.55								
Ile	Gly	Ala	Ser	Leu	Ile	гàг	HIS	TIP								
	_															

# FIGURE 7J

Sequence no. 5: range 1 to 786

Bacillus licheniformis β-lactamase with signal sequence
replaced

	100,00	_										_	0			50
		1	.0			20			30		•	4	•	•		*
	*		*	•		•		*	-	C2.2	-	CAT	GCA	444	CTC	GGG
ATG	AAA	GAT	GAT	TTT	GCA	AAA	CII	GAG	GAA	CAA	111	gvi	Ala	TAR	1-011	Glv
Met	Lys	<b>Asp</b>	Asp	Phe	Ala	Lys	Leu	GIU	Glu	GIR	Phe	VPD	WI W	Lyb	10	10
	•	60			7	10	_		80		_	30		•		*
	•	•		•		•					~~~ <i>b</i>	CCG	TAT	CGG	CCG	GAT
ATC	TIT	GCA	TTG	GAT	ACA	GGT	ACA	AAC	CGG	ACG	37.1	232	Tyr	Ara	Pro	Asp
Ile	Phe	Ala	Leu	Asp	Thr	GIA	Thr	ABII	ALG	1111	447	140	- 7 -	3	150	)
		110			120	)			130	•		140		•	•	
•		•		•				3 000	226	CCT	ጥሞል	ACT	GTA	GGC	GTG	CTT
GAG	CGT	TIT	GCT	TTT	GCT	TCG	ACG	All	AAG	Ala	Teu.	Thr	Val	Glv	Val	Leu
Glu	Arg	Phe	Ala	Phe	WT9	Ser	Thr	116	Lys	A1a		190			200	)
		160	_		170	,	_	7,	30	*		•	•		•	
•		•	•			~~~	Carr	CTC:	AAC	CAG	AGA	ATA	ACA	TAT	ACA	CGT
TTG	CAA	CAG	AAA	TCA	ATA	CAA	GWI	LAN	Asn	Gln	Arg	Ile	Thr	Tyr	Thr	Arg
Leu			Lys	Ser	116	GIU	MBD	230	<i></i>		24	<u>ַ</u>		- 2	250	_
	210	_	_	_	220	•		-1		•	•		•		•	*
*		~~~	~~3	220	TAC	AAC	CCG	ATT	ACG	GAA	AAG	CAC	GTT	GAT	ACG	GGA
GAT	GAT	CIT	GIA	AAC	TAC	vvc	Pro	Tle	Thr	Glu	Lys	His	Val	Asp	Thr	Gly
ABP		Leu	val	270	27+		;	280		_	290			300	)	
	260		•		_	*	Ī	•	•		*		•	•		•
200	ACC.	CTC	277	GAG	CTT	GCG	GAT	GCT	TCG	CTT	CGA	TAT	AGT	GAC	AAT	GCG
Mat	The	LAN	Lva	Glu	Leu	Ala	Asp	Ala	Ser	Leu	Arg	Tyr	Ser	Asp	Asn	Ala
	310	Deu	2,0	320			33	D			340			350		
		•				*	•		*		•	*		*		
GCA	CAG	AAT	CTC	ATT	CIT	AAA	CAA	ATT	GGC	GGA	CCI	GAA	AGT	TIG	AAA	AAG
Ala	Gln	Agn	Leu	Ile	Leu	Lys	Gln	Ile	Gly	Gly	Pro	Glu	Ser	Leu	Lys	Lys
36	0			370		_	380			39	0		•	# DO		
		•		•	*		*		*						CRR	CCA
GAA	CTG	AGG	AAG	ATT	GGT	GAT	GAG	GIT	ACA	AAT	CCC	GAA	CGA	Dhe		CCA
Glu	Leu	Arg	Lys	Ile	Gly	Asp	Glu	Val	Thr	WRIT	Pro	GIU	,— <u>=</u>		GIU	Pro
410		•	42			•	430			440			45	U	•	
•		*	•		•		*				~ · ·	.~	ACT	ACA	CCA	AGA
GAG	TTA	AAT	GAA	GTG	AAT	CCC	GGT	GAA	ACT	CAG	GAT	The	Ser	Thr	Ala	AGA
Glu	Leu	Asn	Glu	Val	Asn	Pro	GIÅ	GIU	THE	GIM	WRD	1111	500			Arg 510
460			470			48	0		•	490	•		200		•	*
•	•		•		•	*			~~	-	CAA	CAT	AAA	CTT	CCA	AGT
GCA	CIT	GTC	ACA	AGC	CIT	CGA	GCC	TIT	3 1 -	LII	G) ··	Agn	Lvs	Leu	Pro	AGT Ser
Ala	Leu	Val	Thr	Ser	Leu	Arg	ATS	Pne	BIA	TIER	GIU	vob	٠,٠			Ser

# FIGURE 7K

									540	3			50			560
		1	520			530		_			•		•	•		•
	•		•	•		•		*			CGA	AAT	ACC	ACT	GJY	GAC
CAA	AAA	CGC	GAG	CII	TTA	ATC	GAT	TGG	ATG	AAA	722	Asn	Thr	Thr	Gly	ASP
Glu	T.VE	Arg	Glu	Leu	Leu	Ile	Asp	IID	met	LY	wa	501	,			610
		-571	D)			,,,,					•	•		-		•
	•	•		•		•	•				·~	~~	AAT	GAC	ATT Lys	GCC
	TTA	ATC	CGT	GCC	GGA	GCG	GCA	TCA	TAT	GGA	ACC	250	212	Acro	TAR	Thr
33-	110	710	ATO	Ala	Gly	Val	Pro	Asp	Gly	TIP	GIU	ATT	M.	ABP	Lys	)
ATE	Den	620	<b>~</b>		630	)		(	640			<b>650</b>		•		
_		620		•			•		•				030	COT	TCG	GAA
	-	BCC	CCG	CCA	AAA	GGA	GAT	CCT	GTC	GGT	GIG	CCG	GAL	991	TGG Ile	110
ATC	ATT	100	CCG	Type	LVA	Glv	ASD	Pro	Val	Gly	Thr	Arg	ABD	Asp	Ile 710	~~
GIA	ALE	Ala	SET	-7-	680	3	-	69	0		•	700	_		710	
	. '	670					•	•		•		•	-			
•		•			_	_	CC)	GTA	TTA	TCC	AGC	AGG	GAT	AAA	AAG	GAC
GTG	GCT	GAT	AAA	ACT	GII	Tan	Ala	val	Leu	Ser	Ser	Arg	Asp	Lys	AAG Lys 760	ASP
Ile	Ile	LLD	Pro	PIO	ATT	Ten	71-	740			75	D _		•	760	
	72	0			730					•	•		•		•	•
•	•		•				2 400	CCA	GAG	GCA	ACA	AAG	GTG	GTA	ATG Met	XXX
GCC	AAG	TAT	GAT	GAT	AAA	CIT	73.0	212	Glu	Ala	Thr	Lys	Val	Val	Met	Lys
_ Ala	Lys	-TYI	qeA .	ASP	TAR	ren	TIE	~-	925			•				
	770			78	0											
	•		•	•		•										
GCC	TTA	AAC	ATG	AAC	GGC	AAA										
Ala	Leu	Asn	Met	A9 (	(											

# INTERNATIONAL SEARCH REPORT

	MACT MI Application No
	PCT/US 96/04059
·	
A. CLASSIFICATION F SUIDECT MATTER	·
C 07 D 471/04.C 07 D 491/04.C 07D 501/1	8.C 12 Q 1/00
According to International Patent Classification (IPC) or to both automal classification and IPC	c6
II. PHILDS SEARCHED	
Minemum documentation scarched (classification system followed by classification symbols)  C 07 D, C 12 Q	
Documentation searched other than missimum documentation to the extent that such document	s are encluded an the ficids scarched
<b>4</b> , 1	·
Electronic data base consisted during the microsponal search (name of data base and, where p	ractical, search terms used)
C. DOCUMENTS CONSIDERED TO BE RELEVANT	Relevant to claim No
Category Catation of document, with indication, where appropriate, of the relevant passage	Refrant to date No.
A EP, A, 0 354 757 (BEECHAM GROUP PLC) 14 February 1990 (14.02.90), claims 1-10.	1-93
A US, A, 5 162 524 (FARINA et al.) 10 November 1992 (10.11.92), claims 1-12.	1-93
A EP, A, 0 053 541 (RHONE-POULENC IND.) 09 June 1982 (09.06.82), ————————————————————————————————————	1-93
Perther documents are tisted in the continuation of box C.	ent family members are listed in asmet.
*Special categories of cited documents:  *A* document defining the general state of the art which is not considered to be of particular relevance  *E* earlier document but published on or after the international filing date  *C* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)  *O* document referring to an oral discharge, use, exhibition or other states  *P* document published prior to the international filing date but later than the priority date claimed  *Date of the actual completion of the international search  *A* document 1996	ent of particular relevance; the claimed invention the considered novel or cannot be considered to an investive step when the document is taken alone tent of particular relevance; the claimed invention the considered to involve an inventive step when the ment is combined with one or more other such docu- auch combination being obvious to a person skilled
Purpose Paint Office, P.B. Stis Patentiam 2	S e.h.

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## ANHANG

### ANNEX

## ANNEXE

zum internationalen Recherchen-bericht über die internationale Patentanmeldung Mr.

to the International Bearch Report to the International Patent Application No.

au rapport de recherche inter-national relatif à la desande de brevet international n°

j.

## PCT/US 96/04059 BAE 130932

In dieses Anhang sind die Mitplieder der Patentfamilien der im obenge- nannten internationalen Recherchenbericht amgeführten Patentfokumente angegeben. Diese Angaben dienen nur zur Unternichtung und erfolgen ohne Bewähr.

This Annex lists the patent family seabers relating to the patent documents cited in the above-mentioned international search report. Ine Uffice is in no way liable for these particulars which are given merely for the purpose of information.

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anger Docu	Rech fehrte itent in sea ment le ra	erchenbericht s Patentdokument document cited nch report de brevet cité pport de recherche	Datum der Veröffentlichung Publication date Date de publication	Mithliedler) der Patent/amilie Patent/amily member(%) Membre(%) de la famille de brevets	Datum der Veröffentlichung Publication date Date de publication	
EP	A2	354757	14-02-90	5F A3 557757 5F A2 557757	27-05-91 36-83-98	
US	À	5162524	i0-11-92	US A 5266691	30-11-93	
EP	Ai	53541	09-04-B2	5-1434007314QY329300:000-01-1410001 90:00400740740747777800-1410001 90:00400740747777800-1410001 90:00400740747777800-1410001 90:0040074074777777777777777777777777777	44415.44415.484.4445.487.487.78777777 	